



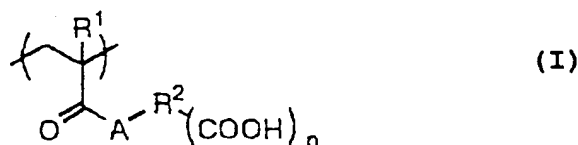
(12) **EUROPEAN PATENT APPLICATION**

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(54) **Negative working photosensitive lithographic printing plate**

(57) A negative working photosensitive lithographic printing plate is disclosed, which comprises a support having thereon at least one photosensitive layer containing a polymeric binder having repeating units represented by formula (I):



wherein R<sup>1</sup> represents a hydrogen atom or a methyl group; R<sup>2</sup> represents a hydrocarbon group which has an alicyclic structure and has 3 to 30 carbon atoms and a valence of n+1; A represents an oxygen atom or -NR<sup>3</sup>-, wherein R<sup>3</sup> represents a hydrogen atom or a monovalent hydrocarbon group having 1 to 10 carbon atoms; and n is an integer of 1 to 5. The negative working photosensitive lithographic printing plate can attain both high productivity and high printing durability. It is especially suitable for drawing with laser light.

**Description**FIELD OF THE INVENTION

5 [0001] The present invention relates to a negative working photosensitive lithographic printing plate (including the printing plate precursor thereof). More particularly, the present invention relates to a negative working photosensitive lithographic printing plate suitable for drawing with a laser light (or beam).

BACKGROUND OF THE INVENTION

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[0002] PS plates comprising a hydrophilic support having thereon an oleophilic photosensitive resin layer have conventionally been widely used as negative working photosensitive lithographic printing plates. In general, a desired printing plate has been obtained from the PS plate by subjecting it to exposure to light (areal exposure) through a mask, i.e., a lith film, and then removing the nonimage areas of the resin layer by dissolution.

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[0003] In recent years, the digitization technology in which image information is electronically processed, stored, and outputted with a computer is spreading widely. With this trend, various new techniques for image output system which are suitable for the digitization technology have come to be practically used. As a result, it is desired to establish a computer-to-plate (CTP) technology in which a photosensitive resin layer is scanned with a highly directional light such as a laser light according to digitized image information to thereby directly produce a printing plate without via the use of a lith film. An important technical subject therefore is to obtain a negative working photosensitive lithographic printing plate suitable for use in that technology.

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[0004] A negative working photosensitive lithographic printing plate capable of such scanning exposure has been proposed which comprises a hydrophilic support having thereon an oleophilic photosensitive resin layer (hereinafter referred to also as "photosensitive layer") containing a photosensitive compound capable of generating an active species, such as radicals or a Brønsted acid, upon exposure to a laser light, and has already been put on the market. From this negative-working photosensitive lithographic printing plate, can be obtained a negative lithographic printing plate by scanning it with a laser light based on digital information to generate an active species, causing the photosensitive layer to undergo a physical or chemical change by the action of the active species to thereby insolubilize the exposed areas of the layer, and subsequently developing the layer.

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[0005] In particular, a negative working photosensitive lithographic printing plate which comprises a hydrophilic support having thereon a photopolymerization type photosensitive layer comprising a photopolymerization initiator having a high photosensitization speed, an ethylenically unsaturated compound capable of undergoing addition polymerization, and a polymeric binder soluble in an alkaline developing solution and which optionally has a protective layer having oxygen barrier properties attains excellent productivity, can be developed easily, and can give a printing plate having desirable printing performances because of its advantages of high resolution and satisfactory ink receptivity.

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[0006] Organic polymers developable with an alkali have conventionally been used as polymeric binders serving as components of photosensitive layers. Examples thereof include the methacrylic acid copolymers, acrylic acid copolymers, itaconic acid copolymers, crotonic acid copolymers, maleic acid copolymers, and partially esterified maleic acid copolymers shown in JP-A-59-44615 (the term "JP-A" as used herein means as "unexamined published Japanese patent application"), JP-B-54-34327 (the term "JP-B" as used herein means an "examined Japanese patent publication"), JP-B-58-12577, JP-B-54-25957, JP-A-54-92723, JP-A-59-53836, JP-A-59-71048, etc.

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[0007] However, the negative working photosensitive lithographic printing plates having a photosensitive layer containing such a conventional polymeric binder have had the following problem. When the speed of scanning is increased in order to further heighten productivity, the exposed areas come not to be sufficiently cured because the exposure light energy applied thereto per unit area becomes lower accordingly. As a result, the image areas are damaged during development by the alkali ingredient contained in the developing solution, so that high printing durability (i.e., high press life) cannot be obtained. Consequently, a further improvement in productivity has been difficult.

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[0008] In JP-A-2000-89460 is disclosed a positive type photosensitive lithographic printing plate containing a vinyl copolymer having structural units comprising a cyclohexane ring and a carboxyl group. However, this technique is clearly different from the present invention.

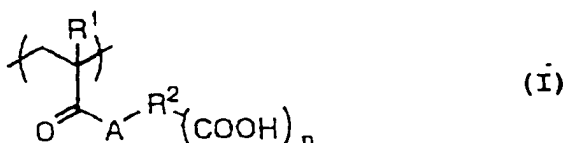
SUMMARY OF THE INVENTION

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[0009] Accordingly, an object of the present invention is to eliminate the above-described drawbacks of the conventional techniques and to provide a negative working photosensitive plate (including the printing plate precursor), especially one suitable for image-drawing with a laser light, from which a lithographic printing plate having sufficient printing durability (i.e.; sufficient press life) can be obtained with high productivity even through exposure conducted under such conditions that the irradiation energy per unit area is low.

[0010] The present inventors made intensive investigations in order to accomplish the object. As a result, it has been found that when a (meth)acrylic acid ester having both of one or more carboxyl groups and an alicyclic hydrocarbon structure is used as a comonomer for imparting developability to a polymeric binder to be used as a component of a photosensitive layer, then the object described above is accomplished. The present invention has been achieved based on this finding.

[0011] The present invention provides a negative working photosensitive lithographic printing plate which comprises a support having thereon at least one photosensitive layer containing a polymeric binder having repeating units represented by formula (I). The term "(meth)acrylic acid" as used herein means acrylic acid and/or methacrylic acid.



[0012] In formula (I),  $\text{R}^1$  represents a hydrogen atom or a methyl group;  $\text{R}^2$  represents a hydrocarbon group which has an alicyclic structure and has 3 to 30 carbon atoms and a valence of  $n+1$ ; A represents an oxygen atom or  $-\text{NR}^3-$ , wherein  $\text{R}^3$  represents a hydrogen atom or a monovalent hydrocarbon group having 1 to 10 carbon atoms; and  $n$  is an integer of 1 to 5.

[0013] One of the features of the negative working photosensitive lithographic printing plate of the present invention is that a (meth)acrylic acid ester having both of one or more carboxyl groups and an alicyclic hydrocarbon structure was used as a comonomer for imparting alkali developability to the polymeric binder. Due to the use of the comonomer, the negative working photosensitive lithographic printing plate of the present invention can give a printing plate having sufficient printing durability with high productivity even through exposure conducted under such conditions that the irradiation energy per unit area is low. Although the reasons for this effect of the present invention have not been elucidated, the following may be thought. During development in conventional techniques, the developing solution penetrates into the image areas wherein the resin should have been sufficiently cured, due to the insufficient irradiation energy in exposure and thus damages the photosensitive layer, resulting in reduced printing durability. In contrast, according to the present invention, the penetration of a developing solution into the image areas is inhibited by incorporating a highly hydrophobic alicyclic hydrocarbon structure in the vicinity of a carboxylic acid to thereby impart high printing durability. Thus, excellent productivity and printing durability can be simultaneously attained by the present invention.

#### DETAILED DESCRIPTION OF THE INVENTION

[0014] The photosensitive layer in the negative working photosensitive lithographic printing plate of the present invention will be explained below in detail.

##### [Polymeric Binder]

[0015] The polymeric binder, which is characteristic of the present invention, is explained first.

[0016] The polymeric binder (referred to also as "binder polymer") used in the photosensitive layer in the negative working photosensitive lithographic printing plate of the present invention is a linear organic high-molecular polymer containing repeating units represented by formula (I) shown above.

[0017] Examples of the group represented by  $\text{R}^2$  in formula (I), which is a hydrocarbon group having an alicyclic structure and has 3 to 30 carbon atoms and a valence of  $n+1$ , include hydrocarbon groups having a valence of  $n+1$  formed by removing  $n+1$  hydrogen atoms from any desired one(s) of the carbon atoms constituting a compound which has an alicyclic structure and may be substituted with one or more any desired substituents. Examples of this compound include cyclopropane, cyclopentane, cyclohexane, cycloheptane, cyclooctane, cyclodecane, dicyclohexyl, tricyclohexyl, norbornane, decahydronaphthalene, perhydrofluorene, tricyclo [5.2.1.0<sup>2,6</sup>] decane, adamantane, quadricyclane, congressane, cubane, spiro[4.4]octane, cyclopentene, cyclohexene, cycloheptene, cyclooctene, cyclodecene, cyclohexadiene, cycloheptadiene, cyclooctadiene, cycloheptatriene, cyclodecatriene, cyclooctatetraene, norbornylene, octahydronaphthalene, bicyclo[2.2.1]heptadiene, bicyclo[4.3.0]nonadiene, dicyclopentadiene, hexahydroanthracene, and spiro[4.5]decadiene.  $\text{R}^2$  has 3 to 30 carbon atoms including those of the substituents.

[0018] In each of such compounds having an alicyclic structure, any one or more of the carbon atoms may have

been replaced with a hetero atom selected from nitrogen, oxygen, and sulfur atoms.

**[0019]** From the standpoint of printing durability,  $R^2$  is preferably a hydrocarbon group which has an optionally substituted alicyclic structure comprising two or more rings and has 5 to 30 carbon atoms and a valence of  $n+1$ , such as e.g., a condensed polycyclic aliphatic hydrocarbon, bridged alicyclic hydrocarbon, spiro aliphatic hydrocarbon, or aliphatic hydrocarbon ring assembly (assembly composed of rings connected to each other through a bond or connecting group). In this case also, the number of carbon atoms includes that of the carbon atoms possessed by any substituent.

**[0020]** Examples of the substituents include monovalent atoms of nonmetallic elements excluding hydrogen and monovalent groups of atoms of nonmetallic elements. Specific examples thereof include halogen atoms (-F, -Br, -Cl, and -I), hydroxyl groups, alkoxy groups, aryloxy groups, mercapto groups, alkylthio groups, arylthio groups, alkyldithio groups, aryldithio groups, amino groups, N-alkylamino groups, N,N-dialkylamino groups, N-arylamino groups, N,N-diarylamino groups, N-alkyl-N-arylamino groups, acyloxy groups, carbamoyloxy groups, N-alkylcarbamoyloxy groups, N-arylcarbamoyloxy groups, N,N-dialkylcarbamoyloxy groups, N,N-diarylcarbamoyloxy groups, N-alkyl-N-arylcarbamoyloxy groups, alkylsulfoxy groups, arylsulfoxy groups, acylthio groups, acylamino groups, N-alkylacylamino groups, N-arylacylamino groups, ureido groups, N'-alkylureido groups, N',N'-dialkylureido groups, N'-arylureido groups, N',N'-diarylureido groups, N'-alkyl-N'-arylureido groups, N-alkylureido groups, N-arylureido groups, N'-alkyl-N-alkylureido groups, N'-alkyl-N-arylureido groups, N',N'-dialkyl-N-alkylureido groups, N',N'-dialkyl-N-arylureido groups, N'-aryl-N-alkylureido groups, N'-aryl-N-arylureido groups, N',N'-diaryl-N-alkylureido groups, N',N'-diaryl-N-arylureido groups, N'-alkyl-N'-aryl-N-alkylureido groups, N'-alkyl-N'-aryl-N-arylureido groups, alkoxycarbonylamino groups, aryloxycarbonylamino groups, N-alkyl-N-alkoxycarbonylamino groups, N-alkyl-N-aryloxycarbonylamino groups, N-aryl-N-alkoxycarbonylamino groups, N-aryl-N-aryloxycarbonylamino groups, formyl groups, acyl groups, carboxyl group and its conjugate base group, alkoxycarbonyl groups, aryloxycarbonyl groups, carbamoyl groups, N-alkylcarbamoyl groups, N,N-dialkylcarbamoyl groups, N-arylcarbamoyl groups, N,N-diarylcarbamoyl groups, N-alkyl-N-arylcarbamoyl groups, alkylsulfinyl groups, arylsulfinyl groups, alkylsulfonyl groups, arylsulfonyl groups, sulfo groups (-SO<sub>3</sub>H) and its conjugate base group, alkoxysulfonyl groups, aryloxysulfonyl groups, sulfinamoyl groups, N-alkylsulfinamoyl groups, N,N-dialkylsulfinamoyl groups, N-arylsulfinamoyl groups, N,N-diarylsulfinamoyl groups, N-alkyl-N-arylsulfinamoyl groups, sulfamoyl, N-alkylsulfamoyl groups, N,N-dialkylsulfamoyl groups, N-arylsulfamoyl groups, N,N-diarylsulfamoyl groups, N-alkyl-N-arylsulfamoyl groups, N-acylsulfamoyl groups and their conjugate base groups, N-alkylsulfonylsulfamoyl groups (-SO<sub>2</sub>NHSO<sub>2</sub>(alkyl)) and their conjugate base groups, N-arylsulfonylsulfamoyl groups (-SO<sub>2</sub>NHSO<sub>2</sub>(aryl)) and their conjugate base groups, N-alkylsulfonylcarbamoyl groups (-CONHSO<sub>2</sub>(alkyl)) and their conjugate base groups, N-arylsulfonylcarbamoyl groups (-CONHSO<sub>2</sub>(aryl)) and their conjugate base groups, alkoxysilyl groups (-Si(O-alkyl)<sub>3</sub>), aryloxysilyl groups (-Si(O-aryl)<sub>3</sub>), hydroxysilyl (-Si(OH)<sub>3</sub>) and its conjugate base group, phosphono (-PO<sub>3</sub>H<sub>2</sub>) and its conjugate base group, dialkylphosphono groups (-PO<sub>3</sub>(alkyl)<sub>2</sub>), diarylphosphono groups (-PO<sub>3</sub>(aryl)<sub>2</sub>), alkylarylphosphono groups (-PO<sub>3</sub>(alkyl)(aryl)), monoalkylphosphono groups (-PO<sub>3</sub>H(alkyl)) and their conjugate base groups, monoarylphosphono groups (-PO<sub>3</sub>H(aryl)) and their conjugate base groups, phosphonooxy (-OPO<sub>3</sub>H<sub>2</sub>) and its conjugate base group, dialkylphosphonooxy groups (-OPO<sub>3</sub>(alkyl)<sub>2</sub>), diarylphosphonooxy groups (-OPO<sub>3</sub>(aryl)<sub>2</sub>), alkylarylphosphonooxy groups (-OPO<sub>3</sub>(alkyl)(aryl)), monoalkylphosphonooxy groups (-OPO<sub>3</sub>H(alkyl)) and their conjugate base groups, monoarylphosphonooxy groups (-OPO<sub>3</sub>H(aryl)) and their conjugate base groups, cyano groups, nitro groups, dialkylboryl groups (-B(alkyl)<sub>2</sub>), diarylboryl groups (-B(aryl)<sub>2</sub>), alkylarylboron groups (-B(alkyl)(aryl)), dihydroxyboryl groups (-B(OH)<sub>2</sub>) and its conjugate base group, alkylhydroxyboryl groups (-B(alkyl)(OH)) and their conjugate base groups, arylhydroxyboryl groups (-B(aryl)(OH)) and their conjugate base groups, aryl groups, alkenyl groups, and alkynyl groups.

**[0021]** Substituents having a hydrogen atom capable of undergoing hydrogen bonding and, in particular, acidic substituents having a lower acid dissociation constant ( $pK_a$ ) than carboxylic acids are undesirable in that such substituents tend to reduce printing durability although this tendency depends on the design of the photosensitive layer. On the other hand, halogen atoms and hydrophobic substituents such as hydrocarbon groups (alkyl groups, aryl groups, alkenyl groups, and alkynyl groups), alkoxy groups, and aryloxy groups are more preferred because these substituents tend to improve printing durability. Especially when the cyclic structure is a six-membered or smaller, monocyclic, aliphatic hydrocarbon such as cyclopentane or cyclohexane, this cyclic structure preferably has such hydrophobic substituents. These substituents may be bonded to each other or to the hydrocarbon group  $R^2$  to form a ring. The substituents may be further substituted.

**[0022]** Examples of the monovalent hydrocarbon group having 1 to 10 carbon atoms represented by  $R^3$  include alkyl groups, aryl groups, alkenyl groups, and alkynyl groups.

**[0023]** Examples of the alkyl groups include linear, branched, and cyclic alkyl groups having 1 to 10 carbon atoms, such as methyl, ethyl, propyl, butyl, pentyl, hexyl, heptyl, octyl, nonyl, decyl, isopropyl, isobutyl, sec-butyl, tert-butyl, isopentyl, neopentyl, 1-methylbutyl, isohexyl, 2-ethylhexyl, 2-methylhexyl, cyclopentyl, cyclohexyl, 1-adamantyl, and 2-norbornyl.

**[0024]** Examples of the aryl groups include ones having 1 to 10 carbon atoms, such as phenyl, naphthyl, and indenyl,

and heteroaryl groups having 1 to 10 carbon atoms and containing one heteroatom selected from the group consisting of nitrogen, oxygen, and sulfur atoms, such as furyl, thienyl, pyrrolyl, pyridyl, and quinolyl.

**[0025]** Examples of the alkenyl groups include linear, branched, and cyclic alkenyl groups having 1 to 10 carbon atoms, such as vinyl, 1-propenyl, 1-butenyl, 1-methyl-1-propenyl, 1-cyclopentenyl, and 1-cyclohexenyl.

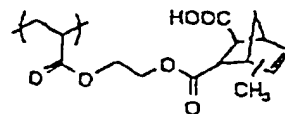
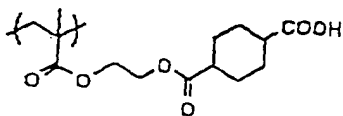
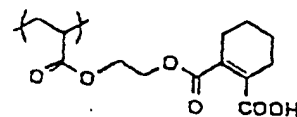
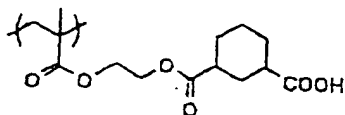
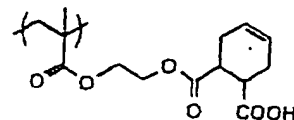
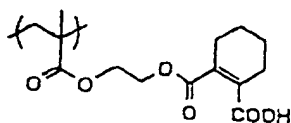
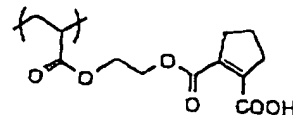
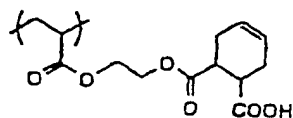
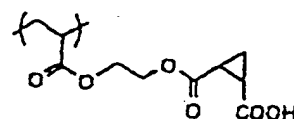
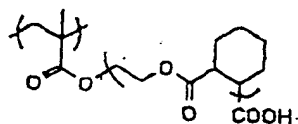
**[0026]** Examples of the alkynyl groups include ones having 1 to 10 carbon atoms, such as ethynyl, 1-propynyl, 1-butylnyl, and 1-octynyl.

**[0027]**  $R^3$  may have one or more substituents. Examples of the substituents include the same substituents as those which may be possessed by  $R^2$ . However, the number of carbon atoms in  $R^3$  including any substituent thereof is from 1 to 10.

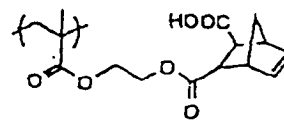
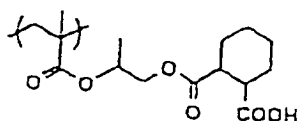
**[0028]** From the standpoint of ease of synthesis, A is preferably an oxygen atom or NH.

**[0029]** Although n is an integer of 1 to 5, it is preferably 1 from the standpoint of printing durability.

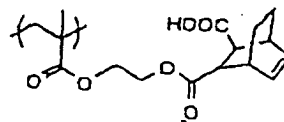
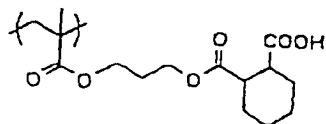
**[0030]** Preferred examples of the repeating units represented by formula (I) are shown below. However, the repeating units in the present invention should not be construed as being limited thereto.



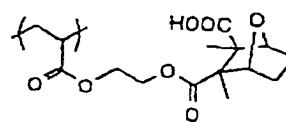
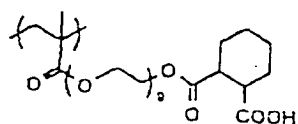
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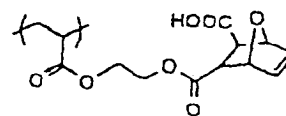
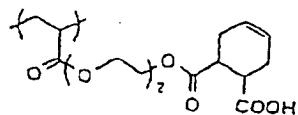
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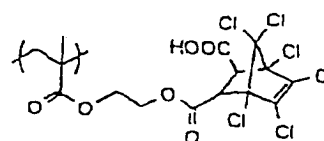
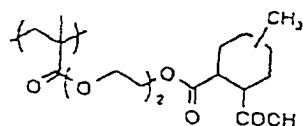


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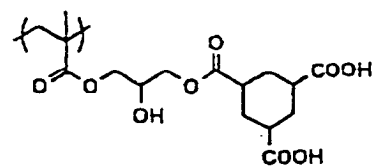
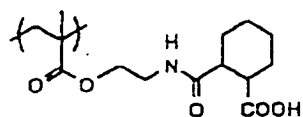
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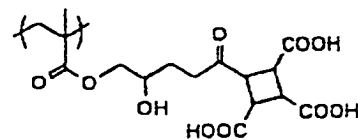
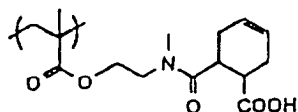


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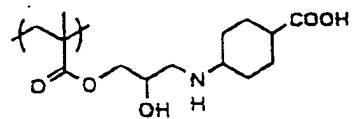
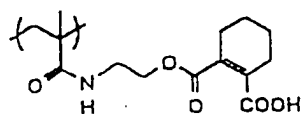


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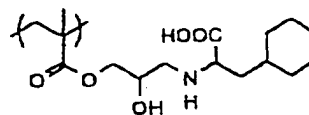
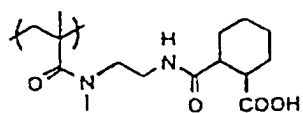


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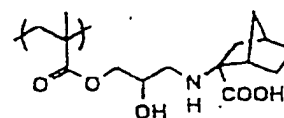
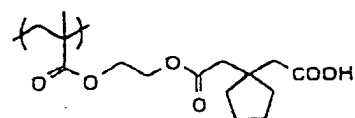
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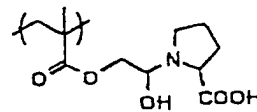
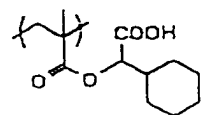


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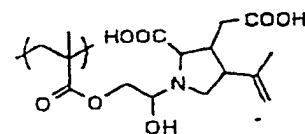
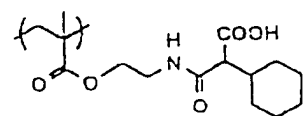
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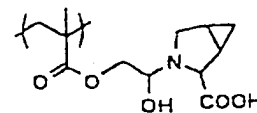
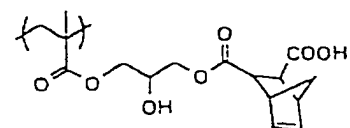


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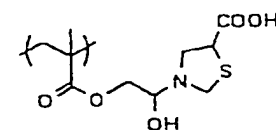
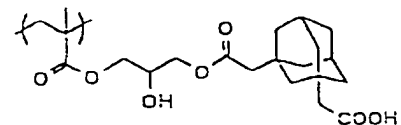


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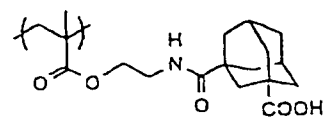
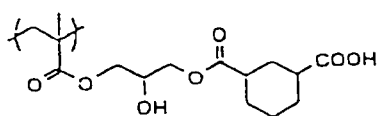
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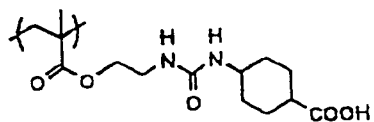


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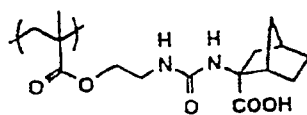
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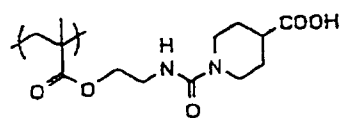


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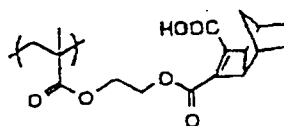


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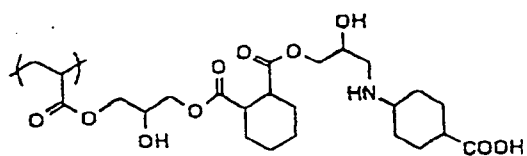


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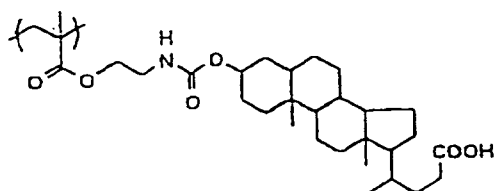
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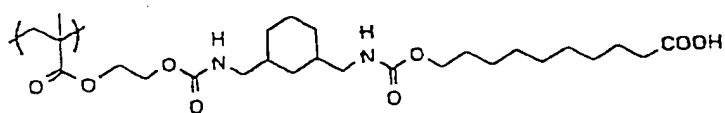
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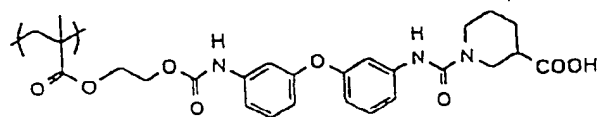
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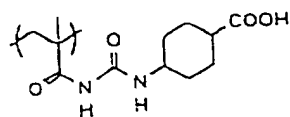


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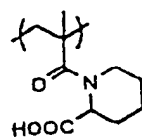
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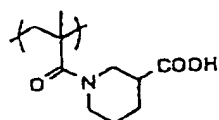
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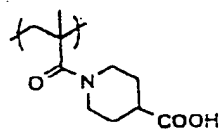
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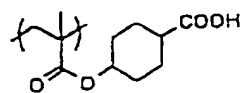


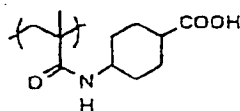
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10 **[0031]** The repeating units represented by formula (I) contained in the binder polymer may be of one kind or two or more kinds.

**[0032]** Although the binder polymer in the present invention may be a polymer consisting only repeating units represented by formula (I), it is usually a copolymer consisting of such repeating units and units of one or more comonomers.

15 **[0033]** In the copolymer, the total content of repeating units represented by formula (I) is suitably determined according to the structure of the copolymer, design of the photosensitive layer, etc. However, the content thereof is preferably from 1 to 99 mol%, more preferably from 10 to 70 mol%, most preferably from 20 to 50 mol%, based on the total amount of all monomer units in the copolymer.

20 **[0034]** In the case where the binder polymer to be used is a copolymer, conventionally well-known monomers capable of radical polymerization can be used as comonomers without particular limitations. Examples thereof include the monomers given in "Kôbunshi Dêta Hando Bukku, Kiso-hen" (edited by The Society of Polymer Science, Japan, Baifukan, 1986). Such comonomers may be used alone or in combination of two or more thereof.

25 **[0035]** The molecular weight of the binder polymer in the present invention is suitably determined from the standpoints of image-forming properties and printing durability. In general, higher molecular weights bring about better printing durability but tend to result in impaired image-forming properties. Conversely, lower molecular weights bring about improved image-forming properties but reduce printing durability. The molecular weight thereof is preferably from 2,000 to 1,000,000, more preferably from 5,000 to 500,000, most preferably from 10,000 to 200,000.

30 **[0036]** In the photosensitive layer in the negative working photosensitive lithographic printing plate of the present invention, the polymeric binder having repeating units represented by formula (I) may be the only polymeric binder used. Alternatively, a mixture of this polymeric binder and one or more other polymeric binders may be used therein. The amount of such polymeric binders which may be optionally used in combination is generally from 1 to 60% by weight, preferably from 1 to 40% by weight, more preferably from 1 to 20% by weight, based on the total amount of all polymeric binders used. Conventionally well-known polymeric binders can be used as the optional polymeric binders without particular limitations. Preferred examples thereof include binders frequently used in the art, such as binders having an acrylic main chain and urethane binders.

35 **[0037]** Although the total amount of the polymeric binder having repeating units represented by formula (I) and one or more optional polymeric binders in the photosensitive layer can be suitably determined, it is generally from 10 to 90% by weight, preferably from 20 to 80% by weight, more preferably from 30 to 70% by weight, based on the total amount of all nonvolatile ingredients in the photosensitive layer.

40 **[0038]** Ingredients which can be incorporated besides the polymeric binders into the photosensitive layer will be explained next.

45 **[0039]** The photosensitive layer in the negative working photosensitive lithographic printing plate of the present invention contains a negative working photosensitive material for image formation. As this photosensitive material, a conventionally well-known one can be used without particular limitations. Examples of the negative working photosensitive material include one comprising a combination of a photo- or heat-acid generator and a compound crosslinkable with an acid catalyst and one comprising a combination of a photo- or heat-polymerization initiator and a compound having an ethylenically unsaturated bond capable of undergoing radical addition polymerization. However, the photosensitive layer is especially preferably a photo- or heat-polymerizable photosensitive layer containing a photo- or heat-polymerization initiator and a compound having an ethylenically unsaturated bond capable of undergoing radical addition polymerization.

50 **[0040]** The negative working photosensitive lithographic printing plate of the present invention is especially suitable for platemaking through direct image-drawing with a laser light having a wavelength of from 300 to 1,200 nm. The printing plate thus produced therefrom has higher printing durability than conventional lithographic printing plates.

[Photo- or Heat-polymerizable Negative working Photosensitive Layer]

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**[0041]** An especially preferred photosensitive layer in the lithographic printing plate of the present invention is a photo- or heat-polymerizable negative working photosensitive layer comprising as essential ingredients a polymeric binder having repeating units represented by formula (I), a photo- or heat-polymerization initiator, and a compound

having at least one ethylenically unsaturated bond capable of undergoing addition polymerization (hereinafter often referred to simply as "addition-polymerizable compound"). Besides these essential ingredients, various additives may be incorporated into this photosensitive layer, such as a co-sensitizer, colorant, plasticizer, and polymerization inhibitor.

5 [Addition-polymerizable Compound]

**[0042]** The addition-polymerizable compound having at least one ethylenically unsaturated double bond to be used in the photo- or heat-polymerizable negative working photosensitive layer is selected from compounds having at least one, preferably at least two ethylenically unsaturated terminal bonds. Such compounds are widely known in this industrial field, and can be used in the present invention without particular limitations. Such compounds are in the chemical forms of, for example, monomer, prepolymers, i.e., dimer, trimer, and oligomer, and mixtures and copolymers thereof. Examples of the monomeric compounds and copolymers thereof include unsaturated carboxylic acids (e.g., acrylic acid, methacrylic acid, itaconic acid, crotonic acid, isocrotonic acid, and maleic acid) and esters and amides thereof. Preferred of these are esters of an unsaturated carboxylic acid with an aliphatic polyhydric alcohol compound and amides of an unsaturated carboxylic acid with an aliphatic polyamine compound. Also preferred are: adducts of an unsaturated carboxylic acid ester or amide having a nucleophilic substituent, e.g., a hydroxyl, amino, or mercapto group, with a mono- or polyfunctional isocyanate or epoxy; and products of the dehydrating condensation reaction of the ester or amide with a mono- or polycarboxylic acid. Preferred examples of the addition-polymerizable compound further include: adducts of an unsaturated carboxylic acid ester or amide having an electrophilic substituent, e.g., an isocyanate or epoxy group, with a mono- or polyfunctional alcohol, amine, or thiol; and products of the substitution reaction of an unsaturated carboxylic acid ester or amide having a releasing substituent, e.g., a halogen or a tosyloxy group, with a mono- or polyfunctional alcohol, amine, or thioether. Other usable examples of the addition-polymerizable compound include compounds formed by the reactions described above wherein an unsaturated phosphonic acid, styrene, vinyl ether, or the like is used in place of the unsaturated carboxylic acid.

**[0043]** Specific examples of the monomeric esters of aliphatic polyhydric alcohol compounds with unsaturated carboxylic acids are as follows. Examples of the acrylic esters include ethylene glycol diacrylate, triethylene glycol diacrylate, 1,3-butanediol diacrylate, tetramethylene glycol diacrylate, propylene glycol diacrylate, neopentyl glycol diacrylate, trimethylolpropane triacrylate, trimethylolpropane tri(acryloyloxypropyl) ether, trimethylolethane triacrylate, hexanediol diacrylate, 1,4-cyclohexanediol diacrylate, tetraethylene glycol diacrylate, pentaerythritol diacrylate, pentaerythritol triacrylate, pentaerythritol tetraacrylate, dipentaerythritol diacrylate, dipentaerythritol hexaacrylate, sorbitol triacrylate, sorbitol tetraacrylate, sorbitol pentaacrylate, sorbitol hexaacrylate, tri(acryloyloxyethyl) isocyanurate, and polyester acrylate oligomers.

**[0044]** Examples of the methacrylic esters include tetramethylene glycol dimethacrylate, triethylene glycol dimethacrylate, neopentyl glycol dimethacrylate, trimethylolpropane trimethacrylate, trimethylolethane trimethacrylate, ethylene glycol dimethacrylate, 1,3-butanediol dimethacrylate, hexanediol dimethacrylate, pentaerythritol dimethacrylate, pentaerythritol trimethacrylate, pentaerythritol tetramethacrylate, dipentaerythritol dimethacrylate, dipentaerythritol hexamethacrylate, sorbitol trimethacrylate, sorbitol tetramethacrylate, bis[p-(3-methacryloxy-2-hydroxypropoxy)phenyl]-dimethylmethane, and bis[p-(methacryloxyethoxy)phenyl]dimethylmethane.

**[0045]** Examples of the itaconic esters include ethylene glycol diitaconate, propylene glycol diitaconate, 1,3-butanediol diitaconate, 1,4-butanediol diitaconate, tetramethylene glycol diitaconate, pentaerythritol diitaconate, and sorbitol tetraitaconate.

**[0046]** Examples of the crotonic esters include ethylene glycol dicrotonate, tetramethylene glycol dicrotonate, pentaerythritol dicrotonate, and sorbitol tetradicrotonate.

**[0047]** Examples of the isocrotonic esters include ethylene glycol diisocrotonate, pentaerythritol diisocrotonate, and sorbitol tetraisocrotonate.

**[0048]** Examples of the maleic esters include ethylene glycol dimaleate, triethylene glycol dimaleate, pentaerythritol dimaleate, and sorbitol tetramaleate.

**[0049]** Other preferred examples of the esters include the aliphatic alcohol esters shown in JP-B-46-27926, JP-B-51-47334, and JP-A-57-196231, the esters having an aromatic basic structure (i.e., skeleton) shown in JP-A-59-5240, JP-A-59-5241, and JP-A-2-226149, and the esters having an amino group shown in JP-A-1-165613.

**[0050]** The ester monomers shown above may be used as a mixture of two or more thereof.

**[0051]** Examples of the monomeric amides of an aliphatic polyamine compound with an unsaturated carboxylic acid include methylenebisacrylamide, methylenebismethacrylamide, 1,6-hexamethylenebisacrylamide, 1,6-hexamethylenebismethacrylamide, diethylenetriaminetrisacrylamide, xylylenebisacrylamide, and xylylenebismethacrylamide.

**[0052]** Examples of other preferred amide monomers include those having a cyclohexylene structure shown in JP-B-54-21726.

**[0053]** Also preferred are addition-polymerizable urethane compounds produced by the addition reaction of an isocyanate with hydroxyl groups. Examples thereof include the vinyl urethane compounds containing two or more polym-

erizable vinyl groups per molecule described in, e.g., JP-B-48-41708. The vinyl urethane compounds are produced by addition reaction of a polyisocyanate compound having two or more isocyanate groups per molecule and a hydroxyl-containing vinyl monomer represented by the following formula (II) :

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wherein  $\text{R}^4$  and  $\text{R}^5$  each represent H or  $\text{CH}_3$ .

10 **[0054]** Other preferred addition-polymerizable urethane compounds include urethane acrylates such as those described in JP-A-51-37193, JP-B-2-32293, and JP-B-2-16765 and the urethane compounds having an ethylene oxide skeleton described in JP-B-58-49860, JP-B-56-17654, JP-B-62-39417, and JP-B-62-39418.

**[0055]** Furthermore, use of the addition-polymerizable compounds having in the molecule an amino structure or sulfide structure described in JP-A-63-277653, JP-A-63-260909, or JP-A-1-105238 is effective in obtaining a photopolymerizable composition having an exceedingly high photosensitization speed.

15 **[0056]** Other examples of the addition-polymerizable compound include polyester acrylates such as those described in JP-A-48-64183, JP-B-49-43191, and JP-B-52-30490 and polyfunctional acrylates or methacrylates such as epoxy acrylates obtained by reacting an epoxy resin with (meth) acrylic acid. The examples further include the specific unsaturated compounds described in JP-B-46-43946, JP-B-1-40337, and JP-B-1-40336 and the vinylphosphonic acid compounds described in JP-A-2-25493. In some cases, the structure having a perfluoroalkyl group described in JP-A-61-22048 is advantageously used. Also usable are the photocurable monomers and oligomers shown in *Nihon Setchaku Kyōkai-shi*, Vol.20, No.7, pp.300-308 (1984).

**[0057]** Details of such addition-polymerizable compounds concerning the structures thereof and the methods of use, e.g., as to whether the addition-polymerizable compounds are used alone or in combination, the amount thereof to be added, etc., can be suitably determined without particular limitations according to the performance design of the negative working photosensitive lithographic printing plate to be finally obtained. For example, selections are made from the following standpoints. From the standpoint of photosensitization speed, structures having a high unsaturated-group content per molecule are preferred. In many cases, addition-polymerizable compounds having a functionality of 2 or higher are preferred. From the standpoint of increasing the strength of the image areas, i.e., the cured film, it is preferred to use an addition-polymerizable compound having a functionality of 3 or higher. A technique effective in regulating both photosensitivity and strength is to use a combination of addition-polymerizable compounds differing in functionality or polymerizable group (e.g., acrylic esters, methacrylic esters, styrene compounds, and vinyl ether compounds). Compounds having a high molecular weight and compounds having high hydrophobicity bring about a high photosensitization speed and excellent film strength but, in some cases, are undesirable from the standpoints of the rate of development and precipitation from a developing solution. Furthermore, the selection of addition-polymerizable compounds and the method of using the same are important factors which influence the compatibility and dispersibility thereof in other ingredients in the photosensitive layer (e.g., the binder polymer, initiator, and colorant). For example, use of a compound with a low polymerization degree or use of a combination of two or more addition-polymerizable compounds can improve compatibility.

25 **[0058]** There also are cases where an addition-polymerizable compound having a specific structure is selected in order to improve adhesion to the support or to the overcoat layer which will be described later, etc. With respect to the proportion of the addition-polymerizable compounds in the photosensitive layer, higher proportions are advantageous from the standpoint of sensitivity. However, too high proportions thereof pose problems such as the occurrence of undesirable phase separation, troubles in production steps due to tackiness of the photosensitive layer (e.g., production failures attributable to the transfer or adhesion of photosensitive-layer components), and precipitation from a developing solution. From these standpoints, the addition-polymerizable compounds are used in an amount of preferably from 5 to 80% by weight, more preferably from 25 to 75% by weight, based on the nonvolatile ingredients in the photosensitive layer. The addition-polymerizable compounds may be used alone or in combination of two or more thereof. Furthermore, addition-polymerizable compounds having an appropriate structure may be selected and incorporated in an appropriate amount without particular limitations from the standpoints of the degree of polymerization inhibition by oxygen, resolution, blurring, change in refractive index, surface tackiness, etc. In some cases, a coating method for forming a layer constitution containing an undercoat layer and an overcoat layer may be employed.

[Photo- or Heat-polymerization Initiator]

55 **[0059]** As the photopolymerization initiator can be used various well-known photopolymerization initiators shown in patent and other literatures. One of these or a combination of two or more of these (photopolymerization initiator system) can be suitably selected according to the wavelength for the light source to be used.

- [0060]** Various photopolymerization initiators (initiator systems) have been proposed for use in the case of using a blue semiconductor laser, an Ar laser, the second higher harmonic (wave) of an infrared semiconductor laser, or an SHG-YAG laser as a light source. Examples thereof include the photo-reducing dyes of some kind shown in, e.g., U. S. Patent 2,850,445, such as Rose Bengal, Eosin, and erythrocine; a system comprising a combination of a dye and an initiator, e.g., a combination of a dye and an amine (see JP-A-44-20189); a system comprising a combination of a hexaarylbiimidazole, a radical generator, and a dye (see JP-A-45-37377); a system comprising a hexaarylbiimidazole and a p-dialkylaminobenzylidene ketone (see JP-B-47-2528 and JP-A-54-155292); a system comprising a cyclic cis- $\alpha$ -dicarbonyl compound and a dye (see JP-A-48-84183); a system comprising a cyclic triazine and a merocyanine dye (see JP-A-54-151024); a system comprising 3-ketocoumarin and an activator (see JP-A-52-112681 and JP-A-58-15503); a system comprising biimidazole, a styrene derivative, and a thiol (see JP-A-59-140203); a system comprising an organic peroxide and a dye (see JP-A-59-1504, JP-A-59-140203, JP-A-59-189340, JP-A-62-174203, JP-B-62-1641, and U.S. Patent 4,766,055); a system comprising a dye and an active halogen compound (see JP-A-63-1718105, JP-A-63-258903, JP-A-3-264771, etc.); a system comprising a dye and a borate compound (see JP-A-62-143044, JP-A-62-150242, JP-A-64-13140, JP-A-64-13141, JP-A-64-13142, JP-A-64-13143, JP-A-64-13144, JP-A-64-17048, JP-A-1-229003, JP-A-1-298348, JP-A-1-138204, etc.); a system comprising a dye having a rhodanine ring and a radical generator (see JP-A-2-179643 and JP-A-2-244050); a system comprising a titanocene and a 3-ketocoumarin dye (see JP-A-63-221110); a system comprising a combination of a titanocene, a xanthene dye, and an addition-polymerizable ethylenically unsaturated compound containing an amino or urethane group (see JP-A-4-221958 and JP-A-4-219756); a system comprising a titanocene and a specific merocyanine dye (see JP-A-6-295061); and a system comprising a titanocene and a dye having a benzopyran ring (see JP-A-8-334897).
- [0061]** Especially preferred photopolymerization initiators (initiator systems) in the photosensitive layer in the negative working photosensitive lithographic printing plate of the present invention contain at least one titanocene.
- [0062]** Any titanocene compound may be used in the photopolymerization initiator (initiator system) in the present invention as long as it is capable of generating an active radical upon irradiation with a light in the presence of a sensitizing dye. One or more titanocene compounds to be used in the present invention may be suitably selected from the well-known titanocene compounds described in, e.g., JP-A-59-152396, JP-A-61-151197, JP-A-63-41483, JP-A-63-41484, JP-A-2-249, JP-A-2-291, JP-A-3-27393, JP-A-3-12403, and JP-A-6-41170.
- [0063]** Specific examples of the titanocene compounds include dicyclopentadienyltitanium dichloride, dicyclopentadienyltitanium-bisphenyl, dicyclopentadienyltitanium-bis-2,3,4,5,6-pentafluorophenyl (hereinafter referred to also as "T-1"), dicyclopentadienyltitanium-bis-2,3,5,6-tetrafluorophenyl, dicyclopentadienyltitanium-bis-2,4,6-trifluorophenyl, dicyclopentadienyltitanium-bis-2,6-difluorophenyl, dicyclopentadienyltitanium-bis-2,4-difluorophenyl, dimethylcyclopentadienyltitanium-bis-2,3,4,5,6-pentafluorophenyl, dimethylcyclopentadienyltitanium-bis-2,3,5,6-tetrafluorophenyl, dimethylcyclopentadienyltitanium-bis-2,4-difluorophenyl, and bis(cyclopentadienyl)bis(2,6-difluoro-3-(pyl-1-yl)phenyl) titanium (hereinafter referred to also as "T-2").
- [0064]** Those titanocene compounds can be chemically modified in various ways in order to improve the properties of the photosensitive layer. Examples of usable modification techniques include bonding to a radical-generating part of, e.g., a sensitizing dye or addition-polymerizable unsaturated compound, incorporation of a hydrophilic group, improvement of compatibility, incorporation of a substituent for inhibiting crystal precipitation, incorporation of a substituent for improving adhesion, and conversion into a polymer.
- [0065]** Methods for using those titanocene compounds also can be suitably determined without particular limitations according to the performance design of the negative working photosensitive lithographic printing plate, as in the case of the addition-polymerizable compounds described above. For example, a titanocene compound can have enhance compatibility in the photosensitive layer when used in combination with one or more other titanocene compounds. Larger amounts of the photopolymerization initiators to be used, including the titanocene compounds, are usually advantageous from the standpoint of photosensitivity. Sufficient photosensitivity is obtained when a photopolymerization initiator is used in an amount of generally from 0.5 to 80 parts by weight, preferably from 1 to 50 parts by weight, per 100 parts by weight of the nonvolatile ingredients in the photosensitive layer. On the other hand, smaller titanocene amounts are preferred from the standpoint of the fogging property caused by a light of around 500 nm as a result of the use of yellow or white light or the like; this fogging property is a major purpose of the present invention. Even when the amount of one or more titanocene compounds is reduced to 6 parts by weight or smaller, especially to 1.9 parts by weight or smaller, and further to 1.4 parts by weight or smaller, sufficient photosensitivity can be obtained by using the titanocene compounds in combination with a sensitizing dye.
- [0066]** In the case where a laser which emits infrared rays having a wavelength of from 760 to 1,200 nm is to be used as a light source, it is generally indispensable to use an infrared absorber. Infrared absorbers function to absorb infrared rays and convert them into heat. A radical generator is pyrolyzed by the resulting heat to generate a radical. The infrared absorber to be used in the present invention is a dye or pigment having an absorption maximum in the wavelength range of from 760 to 1,200 nm.
- [0067]** As the dye can be used commercial dyes and other well-known dyes shown in the literature, e.g., "Senryō

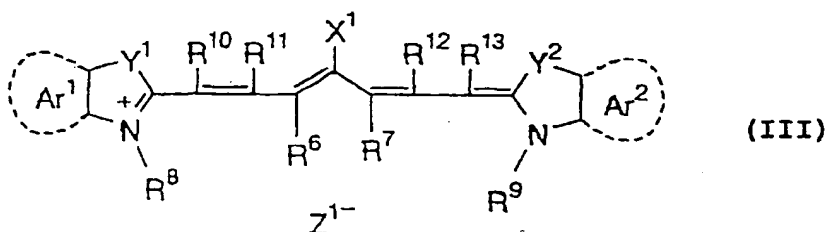
Binran" (edited by Japan Society of Organic Synthesis Chemistry, published in 1970). Examples of the dyes include azo dyes, metal complex azo dyes, pyrazolone azo dyes, naphthoquinone dyes, anthraquinone dyes, phthalocyanine dyes, carbonium dyes, quinoneimine dyes, methine dyes, cyanine dyes, squarylium dyes, pyrylium salts, and metal thiolate complexes.

**[0068]** Preferred examples of the dyes include the cyanine dyes described in, e.g., JP-A-58-125246, JP-A-59-84356, JP-A-59-202829, and JP-A-60-78787; the methine dyes described in, e.g., JP-A-58-173696, JP-A-58-181690, and JP-A-58-194595; the naphthoquinone dyes described in, e.g., JP-A-58-112793, JP-A-58-224793, JP-A-59-48187, JP-A-59-73996, JP-A-60-52940, and JP-A-60-63744; the squarylium dyes described in, e.g., JP-A-58-112792; and the cyanine dyes described in British Patent 434,875.

**[0069]** Other preferred examples thereof include the near-infrared-absorbing sensitizer described in U.S. Patent 5,156,938; the substituted arylbenzo(thio)pyrylium salts described in U.S. Patent 3,881,924; the trimethine thiapyrylium salts described in JP-A-57-142645 (i.e., U.S. Patent 4,327,169); the pyrylium compounds described in JP-A-58-181051, JP-A-58-220143, JP-A-59-41363, JP-A-59-84248, JP-A-59-84249, JP-A-59-146063, and JP-A-59-146061; the cyanine dyes described in JP-A-59-216146; the pentamethine thiopyrylium salts described in U.S. Patent 4,283,475; and the pyrylium compounds disclosed in JP-B-5-13514 and JP-B-5-19702.

**[0070]** Further preferred examples of the dyes include the near-infrared-absorbing dyes represented by formulae (I) and (II) described in the specification of U.S. Patent 4,756,993.

**[0071]** Especially preferred of those dyes are cyanine dyes, squarylium dyes, pyrylium salts, and nickel thiolate complexes. More preferred are cyanine dyes. In particular, the cyanine dyes represented by formula (III) are most preferred.

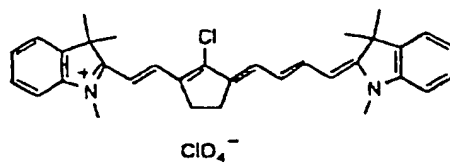


**[0072]** In formula (III),  $X^1$  represents a halogen atom or  $X^2-L^1$ , wherein  $X^2$  represents an oxygen atom or sulfur atom and  $L^1$  represents a hydrocarbon group having 1 to 12 carbon atoms.  $R^6$  and  $R^7$  each independently represent a hydrocarbon group having 1 to 12 carbon atoms. From the standpoint of the storage stability of a coating solution for photosensitive layer,  $R^6$  and  $R^7$  preferably are hydrocarbon groups having 2 or more carbon atoms, and more preferably are bonded to each other to form a five- or six-membered ring.

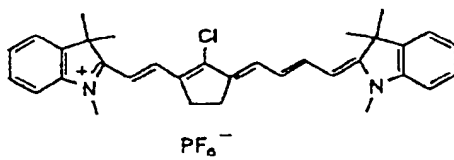
**[0073]**  $Ar^1$  and  $Ar^2$  may be the same or different and each represent an aromatic hydrocarbon group which may have one or more substituents. Preferred examples of the aromatic hydrocarbon group include a benzene ring and a naphthalene ring. Preferred examples of the substituents include hydrocarbon groups having 1 to 12 carbon atoms, halogen atoms, and alkoxy groups having not more than 12 carbon atoms.  $Y^1$  and  $Y^2$  may be the same or different and each represent a sulfur atom or dialkylmethylene group having not more than 12 carbon atoms.  $R^8$  and  $R^9$  may be the same or different and each represent a hydrocarbon group which has not more than 20 carbon atoms and may have one or more substituents. Preferred examples of the substituents include alkoxy groups having not more than 12 carbon atoms, carboxyl, and sulfo.  $R^{10}$ ,  $R^{11}$ ,  $R^{12}$ , and  $R^{13}$  may be the same or different and each represent a hydrogen atom or a hydrocarbon group having not more than 12 carbon atoms. From the standpoint of availability of starting materials,  $R^{10}$  to  $R^{13}$  are preferably hydrogen atoms.  $Z^{1-}$  represents a counter anion, provided that  $Z^{1-}$  need not be present when any of  $R^6$  to  $R^{13}$  has been substituted with a sulfo group. From the standpoint of the storage stability of a coating solution for photosensitive layer,  $Z^{1-}$  is preferably a halogen ion, perchlorate ion, tetrafluoroborate ion, hexafluorophosphate ion, or sulfonate ion, and is more preferably a perchlorate ion, hexafluorophosphate ion, or arylsulfonate ion.

**[0074]** Specific examples of the cyanine dyes represented by formula (III), which can be advantageously used in the present invention, include those shown below. These dyes are given in the specification of JP-A-2001-133969.

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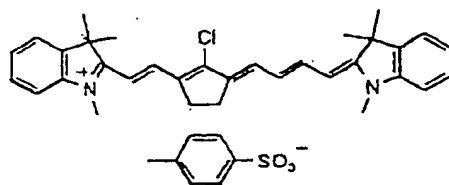


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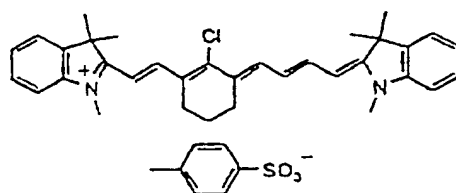
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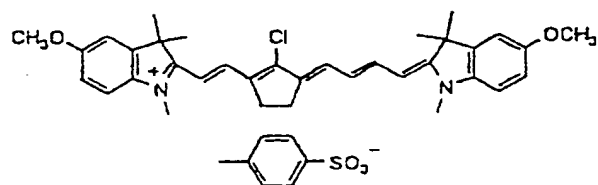
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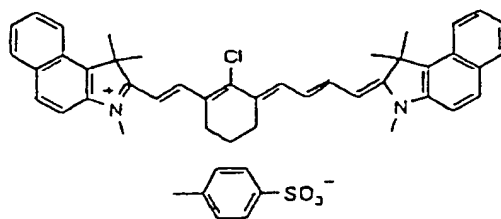
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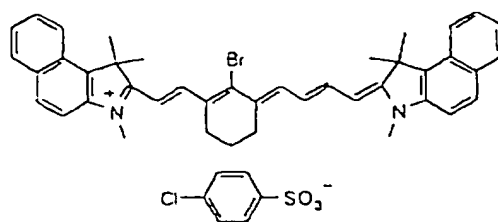
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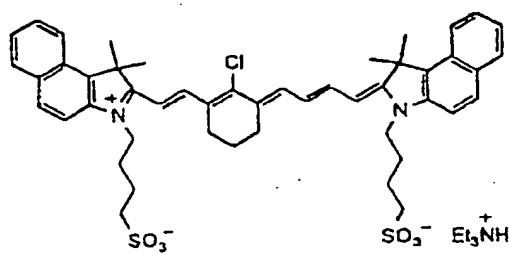
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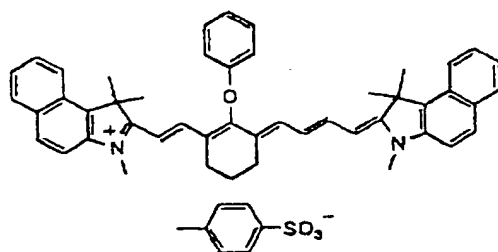
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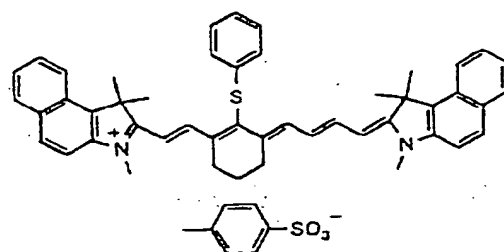
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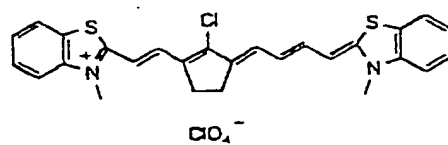
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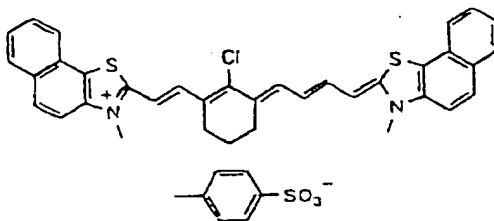
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**[0075]** As the pigment in the present invention can be used commercial pigments and pigments described in a Color Index (C.I.) handbook, "Saishin Ganryō Binran" (edited by Japan Society of Pigment Technology, published in 1977), "Saishin Ganryō Ōyō Gijutsu" (CMC Shuppan, published in 1986), and "Insatsu Inki Gijutsu" (CMC Shuppan, published in 1984).

**[0076]** Examples of the kinds of such pigments include black pigments, yellow pigments, orange pigments, brown pigments, red pigments, purple pigments, blue pigments, green pigments, fluorescent pigments, metal powder pigments, and polymer-bonded pigments. Specific examples of usable pigments include insoluble azo pigments, azo lake pigments, condensed-ring azo pigments, chelate azo pigments, phthalocyanine pigments, anthraquinone pigments, perylene pigments, perinone pigments, thioindigo pigments, quinacridone pigments, dioxazine pigments, isoindolinone pigments, quinophthalone pigments, dyeing lake pigments, azine pigments, nitroso pigments, nitro pigments, natural pigments, fluorescent pigments, inorganic pigments, and carbon black. Preferred of these pigments is carbon black.

**[0077]** Those pigments may be used without undergoing a surface treatment, or may be used after having undergone a surface treatment. Possible methods for surface treatment include a technique in which the surface of a pigment is coated with a resin or wax, a technique in which a surfactant is adhered to the surface of a pigment, and a technique in which a reactive substance (e.g., a silane coupling agent, epoxy compound, or polyisocyanate) is bonded to the surface of a pigment. These methods for surface treatment are described in "Kinzoku Sekken No Seishitsu To Ōyō" (Saiwai Shobo), "Insatsu Inki Gijutsu" (CMC Shuppan, published in 1984), and "Saishin Ganryō Ōyō Gijutsu" (CMC Shuppan, published in 1986).

**[0078]** The particle diameter of the pigment is preferably from 0.01 to 10 μm, more preferably from 0.05 to 1 μm, most preferably from 0.1 to 1 μm. Use of a pigment having a particle diameter of smaller than 0.01 μm is undesirable in that the pigment dispersion shows poor stability in a coating solution for forming an image-forming photosensitive layer. On the other hand, use of a pigment having a particle diameter exceeding 10 μm is undesirable from the standpoint of the evenness of the image-forming photosensitive layer to be formed.

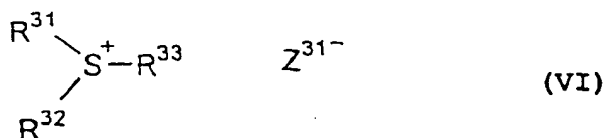
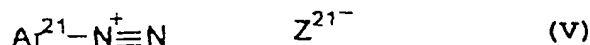
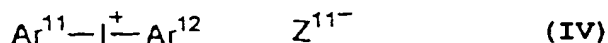
**[0079]** For dispersing the pigment, well-known dispersion techniques for the production of inks, toners, or the like can be used. Examples of usable dispersing machines include an ultrasonic disperser, sand mill, attritor, pearl mill, supermill, ball mill, impeller, disperser, KD mill, colloid mill, dynatron, three-roll mill, and pressure kneader. Details thereof are given in "Saishin Ganryō Ōyō Gijutsu" (CMC Shuppan, published in 1986).

**[0080]** Such infrared absorbers may be incorporated into either the photosensitive layer or another layer, e.g., an overcoat layer or undercoat layer. It is, however, preferred that in the negative working photosensitive lithographic printing plate finally produced, the photosensitive layer should have an optical density of from 0.1 to 3.0 as measured at the wavelength corresponding to the absorption maximum thereof in the wavelength range of from 760 to 1,200 nm. Optical densities thereof outside the range tend to result in reduced sensitivity. Since the optical density of the photosensitive layer is determined by the amount of the infrared absorber added thereto and the thickness of the layer, a desired value of optical density can be obtained by regulating the two factors. The optical density of a photosensitive layer can be measured by an ordinary method. Examples of the method for measurement include: a method in which a photosensitive layer is formed by coating on a transparent or white support in such an amount as to result in a suitably predetermined dry thickness within the range required for lithographic printing plates and is then examined with a transmission type optical densitometer; and a method in which a photosensitive layer is formed on a reflective support such as an aluminum substrate and examined for reflection density.

**[0081]** A heat-decomposable radical generator is used in combination with the infrared absorber. This radical generator means a compound which generates a radical upon irradiation with infrared laser light. Examples of the radical generator include onium salts, triazine compounds having a trihalomethyl group, peroxides, azo polymerization initiators, azide compounds, and quinonediazide. Of these, onium salts are preferred because of their high sensitivity.

**[0082]** Preferred onium salts which can be advantageously used as a radical polymerization initiator in the present invention are explained below. Examples of the preferred onium salts include iodonium salts, diazonium salts, and sulfonium salts. In the present invention, these onium salts function not as an acid generator but as an initiator for radical polymerization. The onium salts which can be advantageously used in the present invention are represented

by formulae (IV) to (VI).

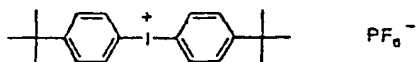


[0083] In formula (IV), Ar<sup>11</sup> and Ar<sup>12</sup> each independently represent an optionally substituted aryl group having not more than 20 carbon atoms (including the carbon atoms of any substituent). In the case where this aryl group has one or more substituents, preferred examples of the substituents include halogen atoms, nitro, alkyl groups having not more than 12 carbon atoms, alkoxy groups having not more than 12 carbon atoms, and aryloxy groups having not more than 12 carbon atoms. Z<sup>11-</sup> represents a counter ion selected from the group consisting of a halogen ion, perchlorate ion, tetrafluoroborate ion, hexafluorophosphate ion, and sulfonate ion, and is preferably a perchlorate ion, hexafluorophosphate ion, or arylsulfonate ion.

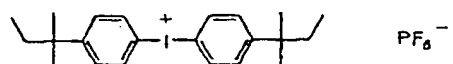
[0084] In formula (V), Ar<sup>21</sup> represents an optionally substituted aryl group having not more than 20 carbon atoms (including the carbon atoms of any substituent). Preferred examples of the substituents include halogen atoms, nitro, alkyl groups having not more than 12 carbon atoms, alkoxy groups having not more than 12 carbon atoms, aryloxy groups having not more than 12 carbon atoms, alkylamino groups having not more than 12 carbon atoms, dialkylamino groups having not more than 12 carbon atoms, arylamino groups having not more than 12 carbon atoms, and diarylamino groups having not more than 12 carbon atoms. Z<sup>21-</sup> represents a counter ion, which has the same meaning as Z<sup>11-</sup>.

[0085] In formula (VI), R<sup>31</sup>, R<sup>32</sup>, and R<sup>33</sup> may be the same or different and each represent an optionally substituted hydrocarbon group having not more than 20 carbon atoms (including the carbon atoms of any substituent). Preferred examples of the substituents include halogen atoms, nitro, alkyl groups having not more than 12 carbon atoms, alkoxy groups having not more than 12 carbon atoms, and aryloxy groups having not more than 12 carbon atoms. Z<sup>31-</sup> represents a counter ion, which has the same meaning as Z<sup>11-</sup>.

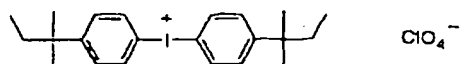
[0086] Specific examples of the onium salts advantageously usable as a radical generator in the present invention include the following onium salts, which are given in the specification of Japanese Patent Application No. 11-310623.



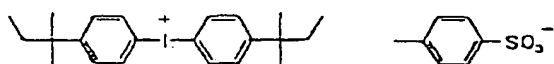
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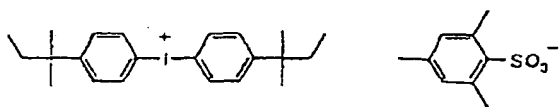
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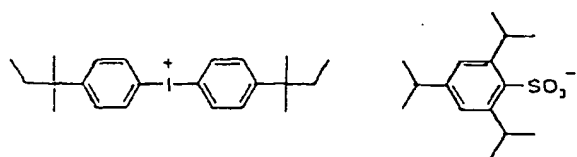
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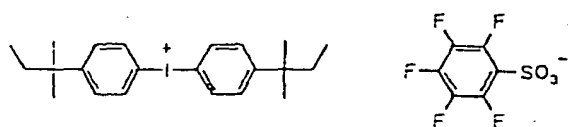
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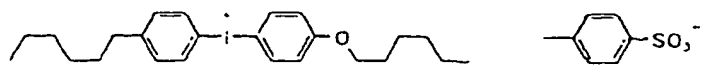
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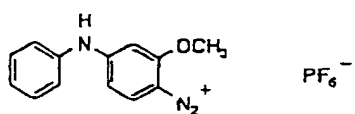
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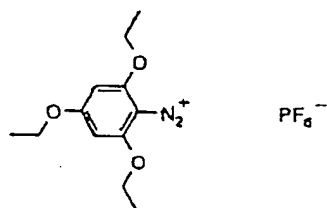


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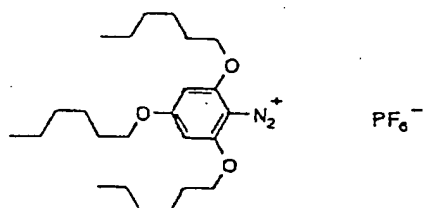


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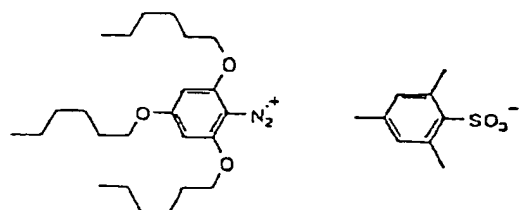
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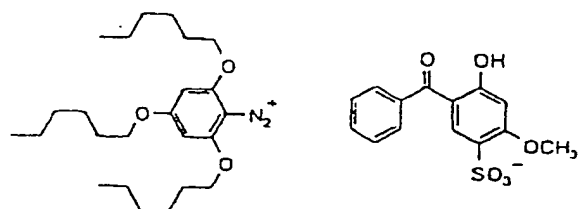
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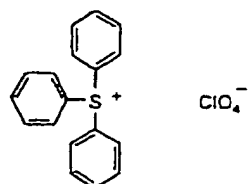
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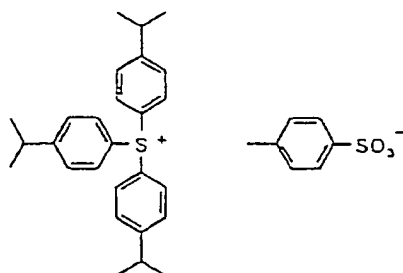
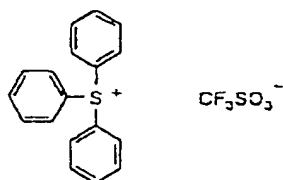
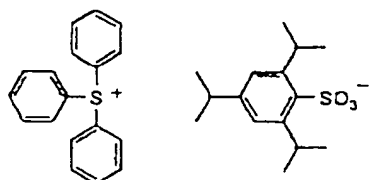
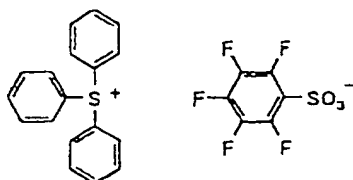


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**[0087]** The heat-decomposable radical generators for use in the present invention have a maximum absorption wavelength of preferably 400 nm or shorter, more preferably 360 nm or shorter. By using such a radical generator having an absorption wavelength within the ultraviolet region, the negative working photosensitive lithographic printing plate can be handled in white light.

**[0088]** Those heat-decomposable radical generators can be added to a coating solution for photosensitive layer in an amount of generally from 0.1 to 50% by weight, preferably from 0.5 to 30% by weight, more preferably from 1 to 20% by weight, based on all the nonvolatile ingredients contained in the coating solution for photosensitive layer. Addition amounts of smaller than 0.1% by weight result in poor sensitivity, while addition amounts of exceeding 50% by weight result in a printing plate in which the nonimage areas are stained during printing. Those radical generators may be used alone or in combination of two or more thereof. The heat-decomposable radical generators may be added to the same layer as the layer for other ingredients or may be added to a layer separately formed. However, it is preferred to add the radical generators to the same layer as the layer for other ingredients.

**[0089]** Besides the basic ingredients described above, other ingredients suitable for the intended use of the photo- or heat-polymerizable negative working photosensitive layer, which is a preferred photosensitive layer for use in the

negative working photosensitive lithographic printing plate of the present invention, or for the intended process for producing the layer, etc. can be suitably added. Preferred examples of such additives will be explained below.

[Co-sensitizer]

**[0090]** The sensitivity of the photopolymerizable photosensitive layer can be further improved by incorporating a co-sensitizer thereinto. Although mechanisms of this improvement are unclear, it is thought that many of these are based on the following chemical process. Namely, it is presumed that a co-sensitizer reacts with various intermediate active species (e.g., radicals, peroxides, oxidizing agents, and reducing agents) which have generated as a result of the photoreactions initiated by light absorption by the photopolymerization initiator (initiator system) described above and during the course of the subsequent addition polymerization reaction to thereby newly yield active radicals. Such co-sensitizers can be roughly divided into (a) co-sensitizers which are reduced to generate active radicals, (b) co-sensitizers which are oxidized to generate active radicals, and (c) co-sensitizers which react with lowly active radicals to convert them into more active radicals or which function as a chain transfer agent. However, with respect to many of the co-sensitizers, there are no common opinions as to which group of these the individual compounds belong.

(a) Compounds which are reduced to generate active radical:

Compounds having a carbon-halogen bond: The compounds are thought to undergo reductional cleavage of the carbon-halogen bond to generate an active radical. Preferred examples thereof include trihalomethyl-s-triazines and trihalomethyloxadiazoles.

Compounds halving a nitrogen-nitrogen bond: The compounds are thought to undergo reductional cleavage of the nitrogen-nitrogen bond to generate an active radical. Preferred examples thereof include hexaarylbimida-zoles.

Compounds having oxygen-oxygen bond: The compounds are thought to undergo reductional cleavage of the oxygen-oxygen bond to generate an active radical. Preferred examples thereof include organic peroxides.

Onium compounds: The compounds are thought to undergo reductional cleavage of a carbon-heteroatom bond or oxygen-nitrogen bond thereof to generate an active radical. Preferred examples thereof include diaryliodonium salts, triarylsulfonium salts, and N-alkoxy-pyridinium (azinium) salts.

Ferrocene and iron-arene complexes: The compounds can reductionally generate an active radical.

(b) Compounds which are oxidized to generate active radical:

Alkylate complexes: The compounds are thought to undergo oxidative cleavage of a carbon-heteroatom bond thereof to generate an active radical. Preferred examples thereof include triarylalkylborates.

Alkylamine compounds: The compounds are thought to undergo oxidative cleavage of a C-X bond on the carbon atom adjacent to the nitrogen atom to generate an active radical. Preferred examples of X include a hydrogen atom, carboxyl, trimethylsilyl, and benzyl. Examples of the compounds include ethanolamine and derivatives thereof, N-phenylglycine and derivatives thereof, and N-trimethylsilylmethylaniline and derivatives thereof.

Sulfur-containing compounds and tin-containing compounds: The compounds having the same structure as the above-described amine compounds except that the nitrogen atom has been replaced with a sulfur or tin atom can generate an active radical by the same mechanism. Furthermore, compounds having an S-S bond are also known to sensitize based on S-S-cleavage.

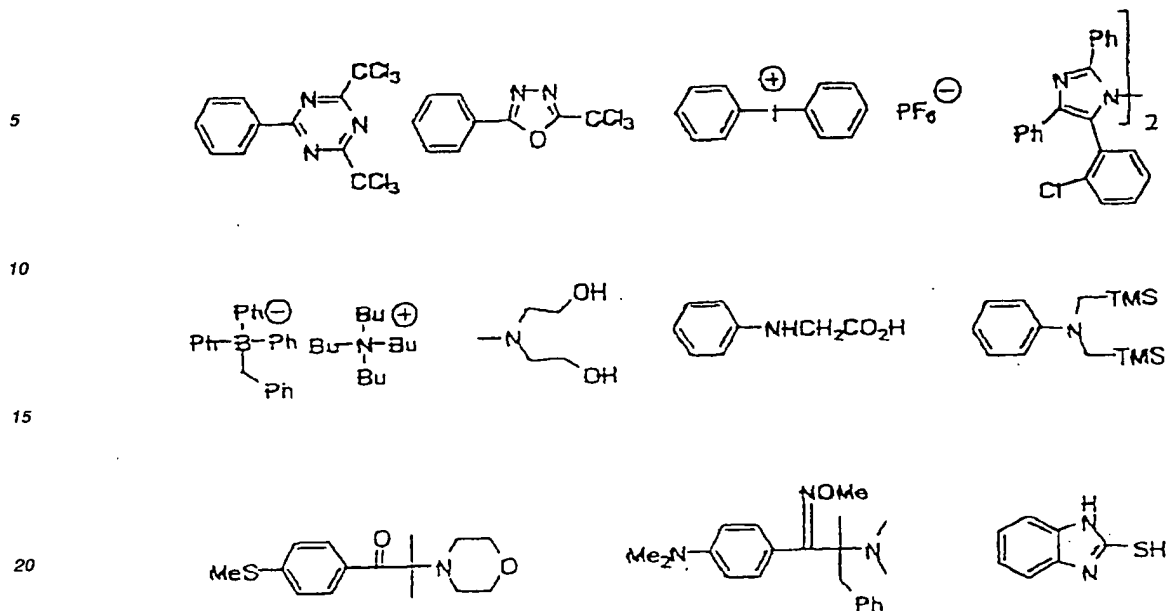
$\alpha$ -Substituted methylcarbonyl compounds: The compounds can undergo oxidative cleavage of the bond between the carbonyl and the  $\alpha$ -carbon to generate an active radical. The compounds formed by converting the carbonyl of these compounds into an oxime ether also function likewise. Examples of such compounds include 2-alkyl-1-[4-(alkylthio)phenyl]-2-morpholinopropane-1-one and the oxime ether compounds obtained by reacting these compounds with a hydroxylamine compound and then etherifying the N-OH.

Specific acid salts: The compounds can reductionally generate an active radical. Examples thereof include sodium arylsulfonates.

(c) Compounds which react with a radical to convert it into a highly active radical or function as a chain transfer agent:

Compounds having, for example, SH, PH, SiH, or GeH in the molecule are used. These compounds can donate hydrogen to a lowly active radical species to generate a radical, or can be oxidized and then deprived of a proton to generate a radical. Examples thereof include 2-mercaptobenzimidazole and derivative thereof.

Many specific examples of those co-sensitizers are given, for example, in JP-A-9-236913 as additives intended to improve sensitivity. Part of these are shown below. However, co-sensitizers usable in the photosensitive layer in the negative working photosensitive lithographic printing plate of the present invention should not be construed as being limited to these.



- 25 **[0091]** Those co-sensitizers also can be subjected to various chemical modifications in order to improve the properties of the photosensitive layer. Examples of usable modification techniques include bonding to a radical-generating part of, e.g., a sensitizing dye, titanocene, or addition-polymerizable unsaturated compound, incorporation of a hydrophilic group, improvement of compatibility, incorporation of a substituent for inhibiting crystal precipitation, incorporation of a substituent for improving adhesion, and conversion into a polymer.
- 30 **[0092]** Those co-sensitizers can be used alone or in combination of two or more thereof. The co-sensitizers are used in an amount of generally from 0.05 to 100 parts by weight, preferably from 1 to 80 parts by weight, more preferably from 3 to 50 parts by weight, per 100 parts by weight of the compound having one or more ethylenically unsaturated double bonds.

35 **[Polymerization Inhibitor]**

- [0093]** In forming the photo- or heat-polymerizable negative working photosensitive layer which is an especially preferred photosensitive layer for use in the negative working photosensitive lithographic printing plate of the present invention, it is desirable to add a small amount of a heat polymerization inhibitor in order to inhibit the compound having one or more, polymerizable, ethylenically unsaturated double bonds from undergoing unnecessary heat polymerization during the production or storage of a negative working photosensitive composition. Appropriate examples of the heat polymerization inhibitor include hydroquinone, p-methoxyphenol, di-t-butyl-p-cresol, pyrogallol, t-butylcatechol, benzoquinone, 4,4'-thiobis(3-methyl-6-t-butylphenol), 2,2'-methylenebis(4-methyl-6-t-butylphenol), and N-nitrosophenylhydroxylamine cerium salt. The heat polymerization inhibitor is added in an amount of preferably about from 0.01 to 5% by weight based on the nonvolatile ingredients contained in the whole composition. Furthermore, a technique for preventing polymerization inhibition caused by oxygen may be used according to need which comprises adding a higher fatty acid or derivative thereof, such as behenic acid or behenamide, and causing the compound to be present in a higher concentration on the surface of the photosensitive layer during the drying of the layer after coating. The amount of the higher fatty acid or derivative to be added is preferably about from 0.5 to 10% by weight based on the nonvolatile ingredients contained in the whole composition.

**[Colorant]**

- 55 **[0094]** A dye or pigment may be added to the photosensitive layer in the negative working photosensitive lithographic printing plate of the present invention for the purpose of coloring the layer. By the addition of a dye or pigment, the printing plate made from the photosensitive lithographic printing plate can be improved in suitability for plate inspection, such as visibility and suitability for examination with an image density meter. Pigments are especially preferred colorants because many dyes reduce the sensitivity of the photopolymerizable photosensitive layer. Examples of the colorant

include pigments such as phthalocyanine pigments, azo pigments, carbon black, and titanium oxide and dyes such as Ethyl Violet, Crystal Violet, azo dyes, anthraquinone dyes, and cyanine dyes. The amount of such dyes and pigments to be added is preferably about from 0.5 to 5% by weight based on the nonvolatile ingredients contained in the whole composition.

[Other Additives]

**[0095]** Other well-known additives may be further added in order to improve the properties of the cured film. Examples thereof include inorganic fillers, plasticizers, and ink receptivity agents capable of improving the ink receptivity of the photosensitive-layer surface.

**[0096]** Examples of the plasticizers include dioctyl phthalate, didodecyl phthalate, triethylene glycol dicaprylate, dimethyl glycol phthalate, tricresyl phosphate, dioctyl adipate, dibutyl sebacate, and triacetyl glycerol. Such plasticizers may be added in an amount of generally not more than 10% by weight based on the sum of the polymeric binder and the addition-polymerizable compound.

**[0097]** For the purpose of improving film strength (printing durability), which will be described later, an additive such as a UV initiator or heat-crosslinking agent may be added in order to heighten the effect of heating/exposure after development.

**[0098]** In a method for forming the photosensitive layer, a photopolymerizable composition comprising the ingredients for the photosensitive layer is dissolved in any of various organic solvents and the resulting coating solution is applied to the interlayer which will be described later. Examples of the solvent used here include acetone, methyl ethyl ketone, cyclohexane, ethyl acetate, ethylene dichloride, tetrahydrofuran, toluene, ethylene glycol monomethyl ether, ethylene glycol monoethyl ether, ethylene glycol dimethyl ether, propylene glycol monomethyl ether, propylene glycol monoethyl ether, acetylacetone, cyclohexanone, diacetone alcohol, ethylene glycol monomethyl ether acetate, ethylene glycol ethyl ether acetate, ethylene glycol monoisopropyl ether, ethylene glycol monobutyl ether acetate, 3-methoxypropanol, methoxymethoxyethanol, diethylene glycol monomethyl ether, diethylene glycol monoethyl ether, diethylene glycol dimethyl ether, diethylene glycol diethyl ether, propylene glycol monomethyl ether acetate, propylene glycol monoethyl ether acetate, 3-methoxypropyl acetate, N,N-dimethylformamide, dimethyl sulfoxide,  $\gamma$ -butyrolactone, methyl lactate, and ethyl lactate. These solvents can be used alone or as a mixture of two or more thereof. An appropriate range of the solid concentration in the coating solution is from 2 to 50% by weight.

**[0099]** The covering amount (i.e., the coating amount) of the coating solution for forming the photosensitive layer mainly influences the sensitivity and developability of the photosensitive layer and the strength and printing durability of the film obtained through exposure. It is therefore desirable to select a suitable covering amount according to the intended use of the photosensitive printing plate. If the covering amount thereof is too low, printing durability is insufficient. On the other hand, too high covering amounts thereof are undesirable in that the photosensitive layer has reduced sensitivity and necessitates much time for exposure and, in addition, the development thereof requires a longer time period. In the case where the negative working photosensitive lithographic printing plate is one intended to be subjected to scanning exposure, which is a major purpose of the present invention, the covering amount of the coating solution for forming the photosensitive layer is preferably about from 0.1 to 10 g/m<sup>2</sup>, more preferably from 0.5 to 5 g/m<sup>2</sup>, on a dry basis.

[Support]

**[0100]** Conventionally well-known hydrophilic supports for use in negative working photosensitive lithographic printing plates can be used without particular limitations as the support of the negative working photosensitive lithographic printing plate of the present invention. It is preferred to use a plate-shaped substrate which is dimensionally stable. Examples thereof include papers, papers laminated with a plastic (e.g., polyethylene, polypropylene, or polystyrene), sheets of metals (e.g., aluminum, zinc, and copper), films of plastics (e.g., cellulose diacetate, cellulose triacetate, cellulose propionate, cellulose butyrate, cellulose acetate butyrate, cellulose nitrate, poly(ethylene terephthalate), polyethylene, polystyrene, polypropylene, polycarbonates, and poly(vinyl acetal), and papers or plastic films coated with any of those metals by laminating or vapor deposition. A surface of such substrates may be subjected to appropriate, well-known, physical or chemical treatments for the purpose of imparting hydrophilicity, improving strength, etc. according to need.

**[0101]** Especially preferred examples of the substrate include papers, polyester films, and aluminum sheets. More preferred of these are aluminum sheets because they have satisfactory dimensional stability, are relatively inexpensive, and can be made to have a surface excellent in hydrophilicity and strength through appropriate surface treatments according to need. Also preferred is a composite sheet comprising a poly(ethylene terephthalate) film and an aluminum sheet bonded thereto, such as that described in JP-B-48-18327.

**[0102]** The term "aluminum sheet" as used herein means a metallic sheet comprising aluminum as a major component



and having dimensional stability. The aluminum sheet is selected from sheets of pure aluminum, sheets of an alloy of aluminum as the main component with a small amount of one or more foreign elements, and plastic films or papers coated with aluminum (or an aluminum alloy) by laminating or vapor deposition.

**[0103]** In the following explanations, the aforementioned substrates comprising aluminum or an aluminum alloy are inclusively referred to as aluminum supports. Examples of elements other than aluminum which are contained in the aluminum alloy include silicon, iron, manganese, copper, magnesium, chromium, zinc, bismuth, nickel, and titanium. The content of such foreign elements in the aluminum alloy is not more than 10% by weight. Although a sheet of pure aluminum is preferred in the present invention, an aluminum sheet containing a slight amount of non-aluminum elements may be used because completely pure aluminum is difficult to produce by the current refining technology. As described above, the aluminum sheet to be used in the present invention is not limited in composition. It can be suitably selected from sheets of conventionally well-known aluminum materials such as, JIS A 1050, JIS A 1100, JIS A 3103, and JIS A 3005. The thickness of the aluminum substrate to be used in the present invention is about from 0.1 to 0.6 mm. This thickness can be suitably varied according to the size of the printing machine to be used, the size of the intended printing plate, and the desire of the user. The aluminum substrate may be suitably subjected according to need, to the surface treatments which will be described below. It may, of course, be used without any surface treatment.

#### [Surface-roughening Treatment]

**[0104]** Examples of surface-roughening treatments include mechanical surface roughening, chemical etching, and electrolytic graining such as those disclosed in JP-A-56-28893. Also usable are an electrochemical surface-roughening method in which the aluminum surface is electrochemically roughened in an electrolytic solution of hydrochloric acid or nitric acid and mechanical surface-roughening methods such as the wire brush graining method in which the aluminum surface is scratched with a metal wire, the ball graining method in which the aluminum surface is grained with abrasive spheres and an abrasive material, and the brush graining method in which the aluminum surface is roughened with a nylon brush and an abrasive material. These surface-roughening methods may be used alone or in combination of two or more thereof.

**[0105]** Of those methods, the electrochemical method in which the aluminum surface is chemically roughened in an electrolytic solution of hydrochloric acid or nitric acid is useful for surface roughening. The quantity of electricity applied to the substrate at the anode is in the range of from 50 to 400 C/dm<sup>2</sup>. More specifically, it is preferred to conduct AC and/or DC electrolysis in an electrolytic solution containing 0.1 to 50% hydrochloric acid or nitric acid under the conditions of a temperature of from 20 to 80°C, an electrolysis time of from 1 second to 30 minutes, and a current density of from 100 to 400 C/dm<sup>2</sup>.

**[0106]** The aluminum substrate which has undergone such a surface-roughening treatment may be chemically etched with an acid or alkali. Examples of etching agents suitable for this etching include caustic soda, sodium carbonate, sodium aluminate, sodium metasilicate, sodium phosphate, potassium hydroxide, and lithium hydroxide. The preferred ranges of concentration and temperature are from 1 to 50% and from 20 to 100°C, respectively. After the etching, pickling is conducted in order to remove the smut remaining on the surface. Examples of acids which can be used for this pickling include nitric acid, sulfuric acid, phosphoric acid, chromic acid, hydrofluoric acid, and borofluoric acid. Preferred examples of methods for smut removal especially after the electrochemical surface-roughening treatment include the method in which the substrate surface is brought into contact with 15 to 65% by weight sulfuric acid at a temperature of from 50 to 90°C as described in JP-A-53-12739 and the method in which the substrate surface is alkali-etched as described in JP-B-48-28123.

**[0107]** Methods and conditions for the surface-roughening treatment and post-treatments are not particularly limited as long as the roughened surface of the substrate thus treated has a center-line average surface roughness Ra (defined in JIS B0601) of from 0.2 to 0.5 μm.

#### [Anodization Treatment]

**[0108]** The aluminum substrate thus treated, which has an oxide layer formed thereon, is subjected to an anodization treatment. In this anodization treatment, an aqueous solution of sulfuric acid, phosphoric acid, oxalic acid, or boric acid/sodium borate or a combination of two or more of solutions of these is used as the main component of an electrolytic bath. This electrolytic solution may, of course, contain at least the ingredients usually contained in the aluminum alloy sheet, an electrode, tap water, underground water, etc. The electrolytic solution may further contain second and third ingredients added thereto. Examples of the second and third ingredients include cations such as ions of metals such as Na, K, Mg, Li, Ca, Ti, Al, V, Cr, Mn, Fe, Co, Ni, Cu, and Zn and ammonium ions and anions such as nitrate ions, carbonate ions, chlorine ions, phosphate ions, fluorine ions, sulfite ions, titanate ions, silicate ions, and borate ions. The concentration of such ions in the electrolytic solution may be about from 0 to 10,000 ppm. There are no particular limitations on the conditions for the anodization treatment. However, the treatment is preferably conducted by DC or

AC electrolysis under the conditions of a concentration of from 30 to 500 g/L, a treating-liquid temperature of from 10 to 70°C, and a current density of from 0.1 to 40 A/m<sup>2</sup>. The thickness of the anodic oxidation layer (or film) thus formed by anodization is from 0.5 to 1.5 µm, preferably from 0.5 to 1.0 µm.

**[0109]** Conditions for the anodization treatment should be selected so that the anodic oxidation layer formed on the substrate thus treated has micropores having a diameter of from 5 to 10 nm and a pore density of from  $8 \times 10^{15}$  to  $2 \times 10^{16}$  pores per m<sup>2</sup>.

**[0110]** For hydrophilizing the substrate surface, various well-known methods can be used. In an especially preferred treatment, the substrate surface is hydrophilized with a silicate, polyvinylphosphonic acid, or the like. A film is formed in an amount of generally from 2 to 40 mg/m<sup>2</sup>, more preferably from 4 to 30 mg/m<sup>2</sup>, in terms of silicon or phosphorus amount.

**[0111]** The amount of the film deposited (or coated) can be determined by fluorescent X-ray spectroscopy.

**[0112]** The hydrophilizing treatment described above can be accomplished by immersing the aluminum substrate having an anodic oxidation layer in an aqueous alkali metal silicate or polyvinylphosphonic acid solution having a concentration of from 1 to 30% by weight, preferably from 2 to 15% by weight, and a pH at 25°C of from 10 to 13. The immersion is conducted, for example, at a temperature of from 15 to 80°C for from 0.5 to 120 seconds.

**[0113]** Examples of the alkali metal silicate used in the hydrophilizing treatment include sodium silicate, potassium silicate, and lithium silicate. A hydroxide may be used for heightening the pH of the aqueous alkali metal silicate solution, and examples thereof include sodium hydroxide, potassium hydroxide, and lithium hydroxide.

**[0114]** A salt of an alkaline earth metal or a salt of a Group IVB metal may be incorporated into the treating solution. Examples of the salt of an alkaline earth metal include water-soluble salts such as nitrates, e.g., calcium nitrate, strontium nitrate, magnesium nitrate, and barium nitrate, sulfates, hydrochlorides, phosphates, acetates, oxalates, and borates. Examples of the salt of a Group IVB metal include titanium tetrachloride, titanium trichloride, potassium fluorotitanate, titanium potassium oxalate, titanium sulfate, titanium tetraiodide, zirconium chloride oxide, zirconium dioxide, zirconium oxychloride, and zirconium tetrachloride.

**[0115]** Such alkaline earth metal salts and Group IVB metal salts can be used alone or in combination of two or more thereof. The amount of such metal salts to be added is preferably from 0.01 to 10% by weight, more preferably from 0.05 to 5.0% by weight.

**[0116]** Silicate electrodeposition such as that described in U.S. Patent 3,658,662 is also effective. Furthermore, a surface treatment comprising a combination of the electrolytic graining disclosed in JP-B-46-27481, JP-A-52-58602, or JP-A-52-30503 with the anodization treatment and hydrophilizing treatment described above is also effective in preparing a substrate.

[Interlayer]

**[0117]** An interlayer may be formed in the negative working photosensitive lithographic printing plate of the present invention for the purposes of improving adhesion between the photosensitive layer and the substrate and improving staining properties. Examples of the interlayer include those described in JP-B-50-7481, JP-A-54-72104, JP-A-59-101651, JP-A-60-149491, JP-A-60-232998, JP-A-3-56177, JP-A-4-282637, JP-A-5-16558, JP-A-5-246171, JP-A-7-159983, JP-A-7-314937, JP-A-8-202025, JP-A-8-320551, JP-A-9-34104, JP-A-9-236911, JP-A-9-269593, JP-A-10-69092, JP-A-10-115931, JP-A-10-161317, JP-A-10-260536, JP-A-10-282682, JP-A-11-84674, JP-A-10-69092, JP-A-10-115931, JP-A-11-38635, JP-A-11-38629, JP-A-10-282645, JP-A-10-301262, JP-A-11-24277, JP-A-11-109641, JP-A-10-319600, JP-A-11-84674, JP-A-11-327152, JP-A-2000-10292, JP-A-2000-235254, JP-A-2000-352824, and EP 1091251A2.

[Protective Layer]

**[0118]** In the negative working photosensitive lithographic printing plate having a photo- or heat-polymerizable negative working photosensitive layer, which is a preferred embodiment of the present invention, it is generally preferred to form a protective layer on the photosensitive layer since the photosensitive printing plate is exposed in the air. The protective layer prevents the low-molecular compounds such as oxygen and basic substances present in the air, which inhibit an image-forming reaction induced in the photosensitive layer by exposure, from coming into the photosensitive layer to thereby enable exposure to be conducted in the air. Consequently, the properties which the protective layer is desired to have are: low permeability to low-molecular compounds including oxygen; the property of not substantially inhibiting transmission of the light to be used for exposure; excellent adhesion to the photosensitive layer; and the ability to be easily removed in a development step after exposure. Various means for imparting such properties to a protective layer have conventionally been taken, and are described in detail in U.S. Patent 3,458,311 and JP-A-55-49729. Preferred examples of materials usable for forming the protective layer include water-soluble polymeric compounds having relatively high crystallinity. Specific well-known examples thereof include poly(vinyl alcohol), poly-

vinylpyrrolidone, acid cellulose derivatives, gelatins, gum arabic, and poly(acrylic acid). However, use of a poly(vinyl alcohol), among these polymers, as the main component gives most satisfactory results with respect to basic properties including oxygen barrier properties and removability by development. The poly(vinyl alcohol) to be used in the protective layer may be one partly substituted with an ester, ether, and acetal, as long as it contains unsubstituted vinyl alcohol units, which gives the necessary oxygen barrier properties and water solubility. For the same reason, a poly (vinyl alcohol) composed of units partly derived from one or more comonomers may be used. Examples of the poly(vinyl alcohol) include ones which have a degree of hydrolysis of from 71 to 100% and a polymerization degree of from 300 to 2,400. Specific examples thereof include PVA-105, PVA-110, PVA-117, PVA-117H, PVA-120, PVA-124, PVA-124H, PVA-CS, PVA-CST, PVA-HC, PVA-203, PVA-204, PVA-205, PVA-210, PVA-217, PVA-220, PVA-224, PVA-217EE, PVA-217E, PVA-220E, PVA-224E, PVA-405, PVA-420, PVA-613, and L-8, all manufactured by Kuraray Co., Ltd.

**[0119]** Not only oxygen barrier properties and removability by development but also fogging, adhesion, and scratch resistance are taken in account in selecting ingredients for the protective layer (kind of PVA and use of additives), the amount of the layer to be formed by coating, etc. In general, a protective layer containing a PVA with a higher degree of hydrolysis (a higher content of unsubstituted vinyl alcohol units) and having a larger thickness has higher oxygen barrier properties and is advantageous in sensitivity. However, excessively enhanced oxygen barrier properties pose problems that an unnecessary polymerization reaction occurs during production or storage before exposure, and that imagewise exposure results in unnecessary fogging or line with enlargement. Furthermore, adhesion to the image areas and scratch resistance also are extremely important for plate handling. Specifically, when a hydrophilic layer comprising a water-soluble polymer is superposed on the oleophilic photosensitive layer, then the hydrophilic layer is easily peeled off due to its insufficient adhesive strength and the peeling parts of the photosensitive layer cause defects such as curing failures due to polymerization inhibition by oxygen. For improving adhesion between the two layers so as to eliminate these problems, various proposals have been made. For example, U.S. Patent 4,072,527 describes a technique for obtaining sufficient adhesion which comprises mixing a hydrophilic polymer consisting mainly of a poly (vinyl alcohol) with from 20 to 60% by weight of acrylic emulsion or water-insoluble vinylpyrrolidone/vinyl acetate copolymer, and superposing the mixture on a photosensitive layer. Any of such well-known techniques can be applied to the protective layer in the present invention. Such methods for forming a protective layer by coating are described in detail in, e.g., U.S. Patent 3,458,311 and JP-A-55-49729.

**[0120]** In a process for making a lithographic printing plate from the negative working photosensitive lithographic printing plate of the present invention, the plate may be wholly heated according to need before and during exposure and during the period of from the exposure to development. This heating accelerates an image-forming reaction in the photosensitive layer to bring about advantages of improved sensitivity, improved printing durability, and stabilized sensitivity. Furthermore, the technique in which the image formed through development is wholly subjected to post-heating or exposure is effective in improving image area strength and printing durability. It is generally preferred that the heating before development be conducted under such mild conditions that the plate is heated at 150°C or lower. Too high temperatures pose problems, for example, that the nonimage areas are fogged. Exceedingly severe conditions are used for the heating after development. Usually, this heating is conducted at a temperature in the range of from 200 to 500°C. Too low temperatures are ineffective in sufficiently heightening the image area strength, while too high temperatures pose problems of deterioration of the support and pyrolysis of the image areas.

**[0121]** For exposing the negative working photosensitive lithographic printing plate of the present invention, well-known methods can be used without particular limitations. It is desirable to use a light source which emits a light having a wavelength of from 300 to 1,200 nm. Preferred examples of the light source include various lasers. The mechanism of exposure may be any of the internal drum exposure, external drum exposure, flat bed exposure, and the like. A highly water-soluble ingredient may be used as a component of the photosensitive layer in the negative working photosensitive lithographic printing plate of the present invention, whereby the photosensitive layer can be made soluble in neutral water and weakly alkaline water. A negative working photosensitive lithographic printing plate having this constitution can be used also in such a manner that it is attached to a printing machine and subjected to on-press exposure and development.

**[0122]** Other examples of the light source for use in exposing the negative working photosensitive lithographic printing plate of the present invention include ultrahigh-pressure, high-pressure, medium-pressure, and low-pressure mercury lamps, chemical lamps, carbon arc lamps, xenon lamps, metal halide lamps, various laser lamps emitting visible or ultraviolet light, fluorescent lamps, tungsten lamps, and sunlight.

**[0123]** The negative working photosensitive lithographic printing plate of the present invention is developed after exposure.

**[0124]** An aqueous alkali solution having a pH of 14 or lower is an especially preferred developing solution for use in the development. More preferred is an aqueous alkali solution having a pH of from 8 to 12 and containing an anionic surfactant. Examples of the alkali include inorganic alkalis such as sodium tertiary phosphate, potassium tertiary phosphate, ammonium tertiary phosphate, sodium secondary phosphate, potassium secondary phosphate, ammonium secondary phosphate, sodium carbonate, potassium carbonate, ammonium carbonate, sodium hydrogen carbonate,

potassium hydrogen carbonate, ammonium hydrogen carbonate, sodium borate, potassium borate, ammonium borate, sodium hydroxide, ammonium hydroxide, potassium hydroxide, and lithium hydroxide. Also usable are organic alkalis such as monomethylamine, dimethylamine, trimethylamine, monoethylamine, diethylamine, triethylamine, monoisopropylamine, diisopropylamine, triisopropylamine, n-butylamine, monoethanolamine, diethanolamine, triethanolamine, 5 monoisopropanolamine, diisopropanolamine, ethyleneimine, ethylenediamine, and pyridine.

[0125] Those alkalis may be used alone or in combination of two or more thereof.

[0126] An anionic surfactant is preferably added to the developing solution for use in developing the negative working photosensitive lithographic printing plate of the present invention in an amount of generally from 1 to 20% by weight, preferably from 3 to 10% by weight. Too small amounts thereof result in impaired developability, while too large amounts 10 thereof produce disadvantageous influences such as the deterioration of the image areas in strength, e.g., wearing resistance.

[0127] Examples of the anionic surfactant include sodium lauryl alcohol sulfate, ammonium lauryl alcohol sulfate, sodium octyl alcohol sulfate, salts of alkylarylsulfonic acids, such as sodium isopropyl naphthalenesulfonate, sodium isobutyl naphthalenesulfonate, the sodium salt of polyoxyethylene glycol mononaphthyl ether sulfate, sodium dodecyl- 15 benzenesulfonate, and sodium m-nitrobenzenesulfonate, sulfuric acid esters of higher alcohols having 8 to 22 carbon atoms, such as disodium alkyl sulfates, phosphoric acid esters of aliphatic alcohols, such as sodium cetyl alcohol phosphate, alkylamide sulfonic acid salts such as  $C_{17}H_{33}CON(CH_3)CH_2CH_2SO_3Na$ , and sulfonic acid salts of dibasic aliphatic ester, such as sodium dioctyl sulfosuccinate and sodium dihexyl sulfosuccinate.

[0128] A slightly water-miscible organic solvent, e.g., benzyl alcohol, can be added to the developing solution according to need. Suitable organic solvents have a solubility in water of about 10% by weight or lower. Preferably, an appropriate organic solvent is selected from ones having a solubility in water of 5% by weight or lower. Examples thereof include 1-phenylethanol, 2-phenylethanol, 3-phenylpropanol, 1,4-phenylbutanol, 2,2-phenylbutanol, 1,2-phe- 20 noxyethanol, 2-benzoyloxyethanol, o-methoxybenzyl alcohol, m-methoxybenzyl alcohol, p-methoxybenzyl alcohol, benzyl alcohol, cyclohexanol, 2-methylcyclohexanol, 4-methylcyclohexanol, and 3-methylcyclohexanol. The content of such an organic solvent is preferably from 1 to 5% by weight based on the total amount of the developing solution in use. The amount of the organic solvent to be used is closely related to the amount of the surfactant to be used. It is preferred to increase the amount of the anionic surfactant as the amount of the organic solvent increases. This is because if a large amount of an organic solvent is added to a developing solution containing a small amount of an anionic surfactant, the organic solvent partly remains undissolved, making it impossible to expect satisfactory devel- 25 opability.

[0129] Additives such as an antifoamer and a hard-water softener can be further added according to need. Examples of the hard-water softener include polyphosphoric acid salts such as  $Na_2P_2O_7$ ,  $Na_5P_3O_{10}$ ,  $Na_3P_3O_9$ ,  $Na_2O_4P(NaO_3P)PO_3Na$ , and Calgon (sodium polymetaphosphate); aminopolycarboxylic acids and derivatives thereof, such as ethylenediaminetetraacetic acid and potassium salts and sodium salts thereof, diethylenetriaminepentaacetic acid and po- 35 tassium salts and sodium salts thereof, triethylenetetraminehexaacetic acid and potassium salts and sodium salts thereof, hydroxyethylethylenediaminetriacetic acid and potassium salts and sodium salts thereof, nitrilotriacetic acid and potassium salts and sodium salts thereof, 1,2-diaminocyclohexanetetraacetic acid and potassium salts and sodium salts thereof, and 1,3-diamino-2-propanoltetraacetic acid and potassium salts and sodium salts thereof; and organic phosphonic acids and derivatives thereof, such as 2-phosphono-1,2,4-butanetricarboxylic acid and potassium salts 40 and sodium salts thereof, 2-phosphono-2,3,4-butanetricarboxylic acid and potassium salts and sodium salts thereof, 1-phosphono-1,2,2-ethanetricarboxylic acid and potassium salts and sodium salts thereof, 1-hydroxyethane-1,1-di-phosphonic acid and potassium salts and sodium salts thereof, and aminotri(methylenephosphonic acid) and potassium salts and sodium salts thereof. The optimal amount of such a hard-water softener to be added varies depending on the hardness and amount of the hard water used. In general, however, a hard-water softener may be incorporated into 45 the developing solution at use in an amount of from 0.01 to 5% by weight, preferably from 0.01 to 0.5% by weight.

[0130] In the case where an automatic processor is used in developing the negative working photosensitive lithographic printing plate, the developing solution fatigues according to the amount of printing plates processed. In this case, the processing ability may be recovered with a replenisher or a fresh developing solution. This replenishment is preferably conducted by the method described in U.S. Patent 4,882,246.

[0131] Also preferred are the developing solutions described in JP-A-50-26601, JP-A-58-54341, JP-B-56-39464, 50 JP-B-56-42860, and JP-B-57-7427.

[0132] The negative working photosensitive lithographic printing plate thus developed is post-treated with washing water, a rinsing solution containing a surfactant etc., and a desensitizing solution containing gum arabic, a starch derivative, etc., in the manner described, e.g., in JP-A-54-8002, JP-A-55-115045, or JP-A-59-58431. Various combi- 55 nations of these treatments can be used for the post-treatment of the negative working photosensitive lithographic printing plate of the present invention.

[0133] The lithographic printing plate obtained through such treatments is mounted on an offset press and used to print many sheets.

[0134] A plate cleaner may be used for removing stains present on the printing plate during printing. As the cleaner is used a conventionally well-known plate cleaner for PS plates. Examples thereof include CL-1, CL-2, CP, CN-4, CN, CG-1, PC-1, SR, and IC (all manufactured by Fuji Photo Film Co., Ltd.).

## 5 EXAMPLE

[0135] The present invention will be explained below in more detail by reference to the following Examples, but the present invention should not be construed as being limited thereto.

## 10 SYNTHESIS EXAMPLE 1

[0136] To acetone (1,000 mL) were added cis-1,2-cyclohexanedicarboxylic anhydride (308.3 g), 2-hydroxyethyl methacrylate (273.3 g), and 4-dimethylaminopyridine (4.9 g). This mixture was heated with refluxing for 5 hours, and the acetone was distilled off under reduced pressure. Thereafter, 1 N hydrochloric acid (500 mL) and ethyl acetate (2,000 mL) were added to the residue to extract it. The resulting organic phase was washed twice with saturated aqueous sodium chloride solution (500 mL), and anhydrous magnesium sulfate (100 g) was then added thereto. This mixture was allowed to stand for 1 hour to dehydrate the organic phase. Subsequently, the magnesium sulfate was removed by filtration and the ethyl acetate was distilled off under reduced pressure to obtain a white solid. The solid was pulverized and then added to water (2,000 mL). After this mixture was stirred for 2 hours, the particles were taken out by filtration and dried. As a result, 2-[2-(methacryloyloxy)ethoxy-carbonyl]cyclohexanecarboxylic acid was obtained as a white solid (518.9 g).

## SYNTHESIS EXAMPLE 2

[0137] A solution of allyl methacrylate (35.8 g), the 2-[2-(methacryloyloxy)ethoxycarbonyl]cyclohexanecarboxylic acid (33.0 g) obtained in Synthesis Example 1, and 2,2'-azobis(2,4-dimethylvaleronitrile) (0.99 g) added in propylene glycol monomethyl ether (310 g) was added dropwise to propylene glycol monomethyl ether (310°C) at 70°C over 2.5 hours in a nitrogen stream. After completion of the addition, the resulting mixture was stirred at 70°C for further 2 hours and then allowed to cool. This solution was poured into water (4,000 mL) kept being vigorously agitated, and the resulting mixture was stirred for further 1 hour. The white particles thus precipitated were taken out by filtration and dried. As a result, the binder polymer (P-1) (66.8 g) shown in Table 1 was obtained. The weight-average molecular weight of this polymer was measured by gel permeation chromatography and found to be 46,000 as calculated in terms of polystyrene. This polymer had an acid value of 1.72 meq/g.

[0138] In the same manner, the binder polymers (P-2) to (P-30) shown in Tables 1 and 2 were obtained.

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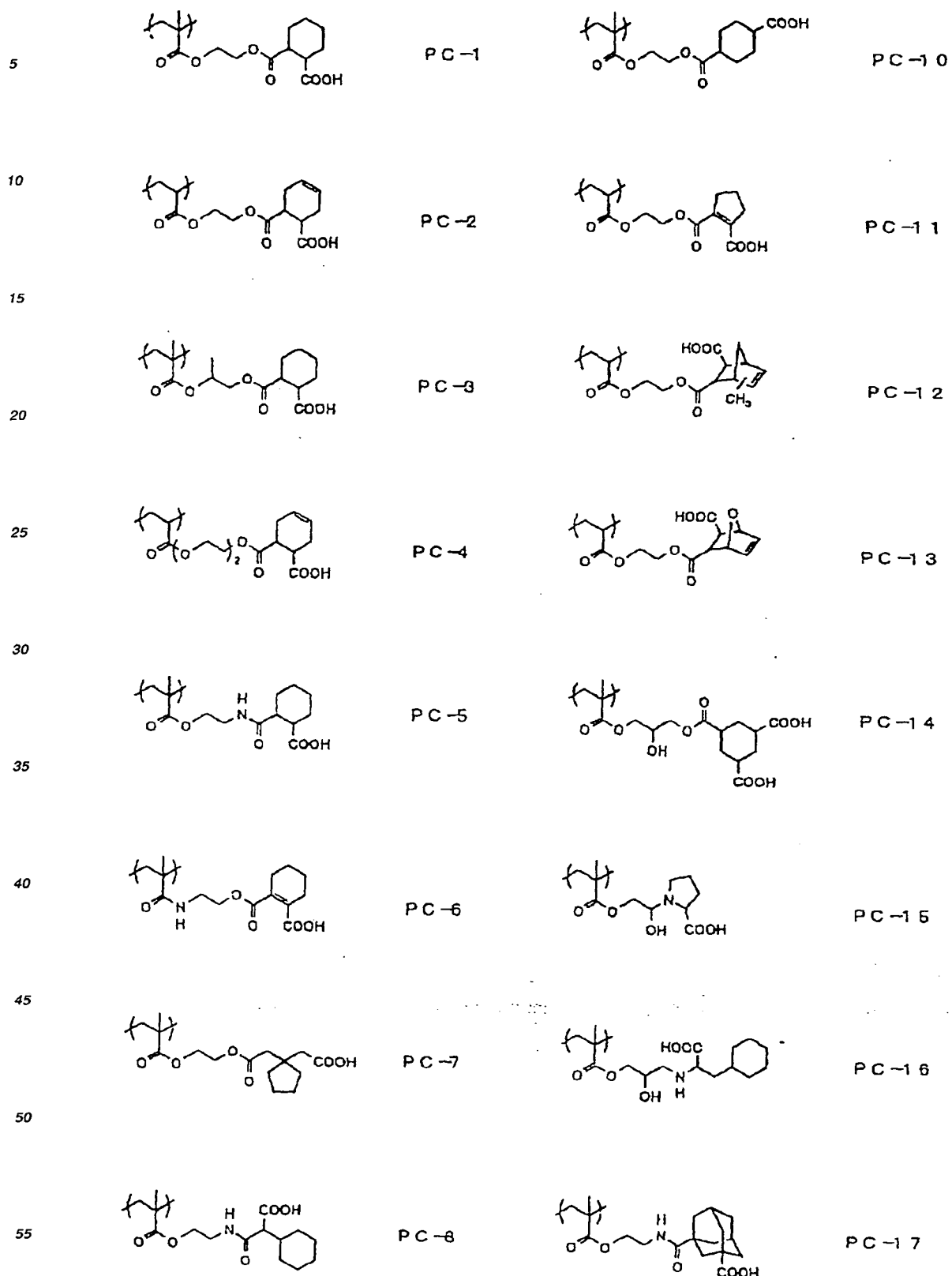
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TABLE 1

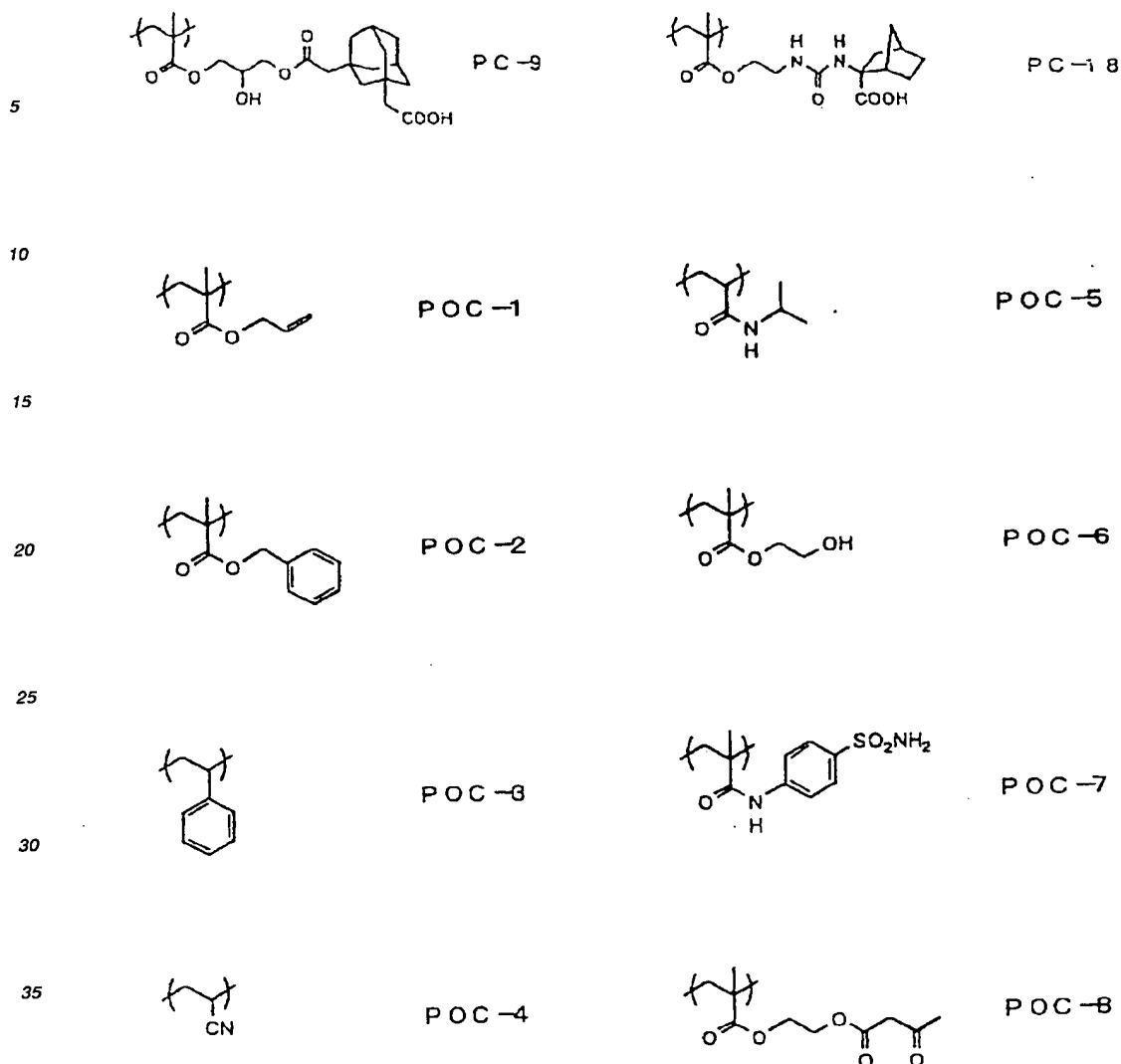
Binder polymer	Repeating units represented by formula (I)	Other Comonomer	Comonomer proportion (in reactor)	Molecular weight (x10000)	Acid value (meq/g)
P-1	PC-1	POC-1	29/71	4.6 0000	1.72
P-2	PC-2	POC-1	35/65	6.2	2.00
P-3	PC-3	POC-1	26/74	3.6	1.50
P-4	PC-3	POC-1	32/68	6.2	1.78
P-5	PC-4	POC-1	28/72	3.7	1.55
P-6	PC-6	POC-1	40/60	11.5	2.10
P-7	PC-8	POC-1	30/70	4.8	1.75
P-8	PC-9	POC-1	15/85	0.9	0.98
P-9	PC-10	POC-1	29/71	4.8	1.72
P-10	PC-12	POC-1	23/77	3.8	1.43
P-11	PC-13	POC-1	27/73	3.9	1.60
P-12	PC-14	POC-1	8/92	2.4	1.15
P-13	PC-16	POC-1	22/78	4.1	1.30
P-14	PC-17	POC-1	30/70	8.4	1.58
P-15	PC-18	POC-1	25/75	4.5	1.41

TABLE 2

Binder polymer	Repeating units represented by formula (I)	Other comonomers		Comonomer proportion (in reactor)	Molecular weight (x10000)	Acid value (meq/g)
P-16	PC-1	POC-1	POC-2	29/51/20	4.6	1.72
P-17	PC-2	POC-2	POC-3	32/50/18	5.2	1.69
P-18	PC-2	POC-3	POC-4	19/45/36	6.2	1.74
P-19	PC-3	POC-2	POC-5	20/40/40	0.9	1.13
P-20	PC-4	POC-5	POC-6	32/50/18	4.5	1.74
P-21	PC-5	POC-1	POC-7	40/43/17	12.3	2.70
P-22	PC-7	POC-1	POC-8	34/55/11	9.7	2.22
P-23	PC-9	POC-2	POC-4	30/50/20	2.0	1.34
P-24	PC-11	POC-2	POC-8	18/60/22	6.5	1.89
P-25	PC-12	POC-5	POC-6	27/53/20	3.9	1.63
P-26	PC-13	POC-3	POC-8	15/65/20	6.3	2.17
P-27	PC-15	POC-2	POC-7	23/58/19	4.4	2.07
P-28	PC-17	POC-1	POC-7	33/55/12	5.6	2.12
P-29	PC-17	POC-5	POC-8	30/60/10	6.1	2.01
P-30	PC-18	POC-4	POC-7	25/40/35	7.6	3.22







#### EXAMPLES 1 TO 20 AND COMPARATIVE EXAMPLES 1 TO 3

**[0139]** Negative working photosensitive lithographic printing plates were produced and evaluated for printing performances in the following manners. The results obtained are shown in Table 3.

#### [Production of Support]

**[0140]** A JIS A 1050 aluminum sheet having a thickness of 0.24 mm and a width of 1030 mm was continuously treated in the following manner.

(a) An existing mechanically surface-roughening apparatus was used to mechanically roughen a surface of the aluminum sheet with rotating nylon brush rollers while supplying a suspension of an abrasive material (pumice) having a specific gravity of 1.12 in water as an abrasive slurry to the aluminum sheet surface. The abrasive material had an average particle diameter of from 40 to 45  $\mu\text{m}$  and a maximum diameter of 200  $\mu\text{m}$ . The nylon brushes were made of nylon-6,10 and had a bristle length of 50 mm and a bristle diameter of 0.3 mm. Three rotating brush rollers were used which each comprised a perforated stainless-steel cylinder having a diameter of 300 mm and bundles of such nylon bristles densely attached thereto by filling them into the perforations. The apparatus had under the brush rollers two supporting rollers (diameter, 200 mm) apart from each other at a distance of 300 mm. The brush rollers were pressed against the aluminum sheet in such a degree that the load imposed on the driving

motor rotating the brush rollers increased to a value higher by 7 kW than that as measured before the brush rollers were pressed against the aluminum sheet. The direction of rotation of the brush rollers was the same as the direction of running of the aluminum sheet, and the rotational speed thereof was 200 rpm.

(b) The aluminum sheet was etched by spraying with an etching solution having a caustic soda concentration of 2.6% by weight and an aluminum ion concentration of 6.5% by weight at a temperature of 70°C to dissolve away a surface layer of the aluminum sheet in an amount of 13 g/m<sup>2</sup>. Thereafter, the aluminum sheet was washed with water by spraying.

(c) The aluminum sheet was subjected to a de-smutting treatment by spraying with an aqueous solution having a nitric acid concentration of 1% by weight (containing 0.5% by weight of aluminum ions) and a temperature of 30°C. Thereafter, the aluminum sheet was washed with water by spraying. The aqueous nitric acid solution used for the de-smutting was a waste liquid resulting from the step of electrochemical surface roughening with an alternating current in an aqueous nitric acid solution.

(d) A 60 Hz AC voltage was used to continuously conduct an electrochemical surface-roughening treatment. The electrolytic solution used was 1% by weight aqueous nitric acid solution (containing 0.5% by weight of aluminum ions and 0.007% by weight of ammonium ions) having a temperature of 40°C. The AC power source used was one providing a trapezoidal rectangular wave alternating current wherein the TP, which is the time required for the current value to increase from zero to a peak, was 2 msec and the duty ratio was 1:1. A carbon electrode was used as a counter electrode to conduct the electrochemical surface-roughening treatment using ferrite as an auxiliary anode. The current density was 30 A/dm<sup>2</sup> in terms of peak value, and the quantity of electricity was 255 C/cm<sup>2</sup> in terms of the sum of electricity at the time when the aluminum sheet was functioning as an anode. To the auxiliary anode was supplied 5% of the current flowing from the power source. After this surface-roughening treatment, the aluminum sheet was washed with water by spraying.

(e) The aluminum sheet was etched by spraying with an etching solution having a caustic soda concentration of 26% by weight and an aluminum ion concentration of 6.5% by weight at 32°C to dissolve away a surface layer of the aluminum sheet in an amount of 0.2 g/m<sup>2</sup>. Thus, the smut ingredients consisting mainly of aluminum hydroxide yielded by the preceding step of electrochemical surface roughening with an alternating current were removed and, simultaneously therewith, the edges of the pits formed were partly dissolved away and rounded to be smooth. Thereafter, the aluminum sheet was washed with water by spraying.

(f) The aluminum sheet was subjected to a de-smutting treatment by spraying with an aqueous solution having a sulfuric acid concentration of 25% by weight (containing 0.5% by weight aluminum ions) and a temperature of 60°C. Thereafter, the aluminum sheet was washed with water by spraying.

(g) An existing anodizing apparatus based on the two-stage-feed electrolytic processing method (lengths of first and second electrolysis zones, 6 m each; length of first feed zone, 3 m; length of second feed zone, 3 m; lengths of first and second feeder electrodes, 2.4 m each) was used to conduct an anodization treatment under the conditions of a sulfuric acid concentration in the electrolysis zones of 170 g/L (containing 0.5% by weight aluminum ions) and a temperature of 38°C. Thereafter, the aluminum sheet was washed with water by spraying. In this anodizing apparatus, a current supplied from a power source flowed to a first feeder electrode disposed in the first feed zone and then to the aluminum sheet through the electrolytic solution to form an oxide film on the surface of the aluminum sheet in the first electrolysis zone. The current then passed through an electrolysis electrode disposed in the first feed zone and returned to the power source. On the other hand, another current supplied from the power source flowed to a second feeder electrode disposed in the second feed zone and likewise to the aluminum sheet through the electrolytic solution to form an oxide film on the surface of the aluminum sheet in the second electrolysis zone. The quantity of electricity fed to the first feed zone from the power source was equal to that fed to the second feed zone from the power source. The current density in the surface of the oxide film in the second feed zone was about 25 A/dm<sup>2</sup>. In the second feed zone, electricity was fed through the oxide film, which weighed 1.35 g/m<sup>2</sup>. The amount of the oxide film finally obtained was 2.7 g/m<sup>2</sup>. The aluminum support obtained through the steps described above is referred to as [AS-1].

#### [Hydrophilizing Treatment]

[0141] The aluminum support [AS-1] was treated with a silicate in order to enhance hydrophilicity, required of the nonimage areas of a printing plate. In this treatment, the aluminum web was passed through a 1.5% aqueous solution of #3 sodium silicate (i.e., disodium trisilicate) kept at 70°C, in such a manner that the web/solution contact time was 15 seconds. Thereafter, the web was washed with water. As a result, silicon was deposited in an amount of 10 mg/m<sup>2</sup>. This support is referred to as [AS-2].

## [Formation of Interlayer]

[0142] A liquid composition (sol) was prepared by the SG method in the following manner. The respective amounts of the following ingredients were weighed in a breaker. This mixture was stirred at 25°C for 20 minutes.

Tetraethoxysilane	38 g
3-Methacryloxypropyltrimethoxysilane	13 g
85% Aqueous phosphoric acid solution	12 g
Ion-exchanged water	15 g
Methanol	100 g

[0143] The resulting solution was transferred to a three-necked flask. A reflux condenser was attached thereto, and this three-necked flask was partly immersed in a room-temperature oil bath. The temperature of the oil bath was elevated to 50°C over 30 minutes while stirring the contents of the three-necked flask with a magnetic stirrer. The mixture was allowed to react for further 1 hour while keeping the bath temperature at 50°C. Thus, a liquid composition (sol) was obtained. This sol was diluted with a methanol/ethylene glycol (20/1 by weight) mixture to obtain a concentration of 0.5% by weight. The dilute sol was applied to the aluminum support [AS-1] with a whirler and then dried at 100°C for 1 minute. Thus, a coated layer was formed in an amount of 3.5 mg/m<sup>2</sup>. This coating amount was the amount of silicon element determined by fluorescent X-ray spectroscopy. The support thus produced is referred to as [AS-3].

[0144] Subsequently, a solution having the following composition was applied to the aluminum support [AS-2] with a wire-bar. The coating was dried with a hot-air drying oven at 90°C for 30 seconds. The amount of the coating was 10 mg/m<sup>2</sup> on a dry basis.

Copolymer of ethyl methacrylate and sodium 2-acrylamido-2-methyl-1-propanesulfonate (75:15 by mol)	0.1 g
2-Aminoethylphosphonic acid	0.1 g
Methanol	50 g
Ion-exchanged water	50 g

[0145] The support thus produced is referred to as [AS-4].

## [Formation of Photosensitive Layer]

[0146] A negative working photosensitive composition having the following makeup was applied to each of the aluminum sheets thus treated, in the dry coating amount shown in Table 3. The coating was dried at 100°C for 1 minute to form a photosensitive layer.

(Negative working Photosensitive Composition)	
Addition-polymerizable compound (compound shown in Table 3)	1.5 g
Binder polymer (compound shown in Table 1)	2.0 g
Sensitizing dye (compound shown in Table 3)	0.2 g
Photopolymerization initiator (compound shown in Table 3)	0.4 g
Co-sensitizing dye (compound shown in Table 3)	0.4 g
Fluorine-based nonionic surfactant (Megafac F-177, manufactured by Dainippon Ink & Chemicals Inc.)	0.03 g
Heat-polymerization inhibitor (N-nitrosophenylhydroxylamine aluminum salt)	0.01 g
Coloring-pigment dispersion having the following composition	2.0 g
Methyl ethyl ketone	20.0 g
Propylene glycol monomethyl ether	20.0 g

(Composition of Coloring-pigment Dispersion)	
Pigment Blue 15:6	15 weight parts
Allyl methacrylate/methacrylic acid copolymer (comonomer proportion, 80/20; weight-average molecular weight, 40,000)	10 weight parts

(continued)

(Composition of Coloring-pigment Dispersion)	
Cyclohexanone	15 weight parts
Methoxypropyl acetate	20 weight parts
Propylene glycol monomethyl ether	40 weight parts

## [Formation of Protective Layer]

**[0147]** A 3% by weight aqueous solution of a poly (vinyl alcohol) (degree of saponification, 98 mol%; degree of polymerization, 550) was applied to the photosensitive layer in an amount of 2 g/m<sup>2</sup> in terms of dry coating amount. The coating was dried at 100°C for 2 minutes.

## [Exposure of Negative working Photosensitive Lithographic Printing Plates]

**[0148]** Using a FD-YAG (532 nm) laser exposure apparatus (plate setter Gutenberg, manufactured by Hidelberg), the negative working photosensitive lithographic printing plates thus obtained were subjected to solid-image exposure and dot image exposure, which was conducted under the conditions of 2,540 dpi, 175 lines per inch, and dot percents increasing from 1 to 99% by percents, while regulating the exposure power so as to result in an exposure energy density of 200  $\mu$ J/cm<sup>2</sup> as measured on the plate surface.

## [Development/Platemaking]

**[0149]** Into automatic processor FLP-813, manufactured by Fuji Photo Film Co., Ltd., were introduced the developing solution shown in Table 3 and finisher FP-2W, manufactured by Fuji Photo Film Co., Ltd. Each plate which had been exposed was developed under the conditions of a developing solution temperature of 30°C and a developing time of 18 seconds. Thus, lithographic printing plates were obtained.

## [Image Area Printing Durability Test]

**[0150]** Printing machine R201, manufactured by Rolland, was used together with Graph G (N), manufactured by Dainippon Ink & Chemicals Inc., as an ink. The solid-image part of each printed matter was examined. The printing durability of the image areas was evaluated in terms of the number of printed sheets obtained by the time when the image began to be thinned. The larger the number, the better the printing durability.

## [Forced Dot Area Printing Durability Test]

**[0151]** Printing machine R201, manufactured by Rolland, was used together with Graph G (N), manufactured by Dainippon Ink & Chemicals Inc., as an ink. At the time when 5,000 sheets had been printed since printing initiation, the dot areas were wiped with a printing sponge impregnated with PS plate cleaner CL-2, manufactured by Fuji Photo Film Co., Ltd., to remove the ink adherent to the plate surface. Thereafter, 10,000 sheets were printed, and the printed matters obtained were visually examined for slipping of image in the dot images.

EXAMPLES 21 TO 30 AND COMPARATIVE EXAMPLES 4 TO 6

**[0152]** The negative working photosensitive lithographic printing plates shown in Table 4 were evaluated for performances in the same manner as in Examples 1 to 20.

**[0153]** The compounds given in Tables 3 and 4 are as follows.

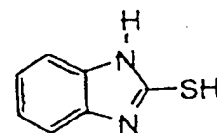
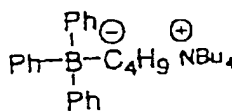
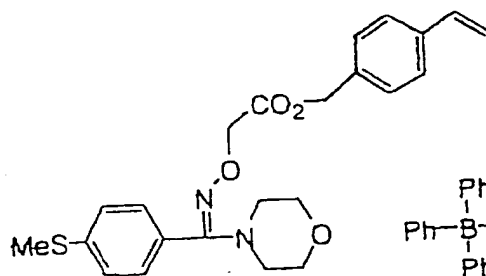
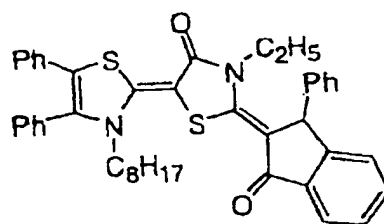
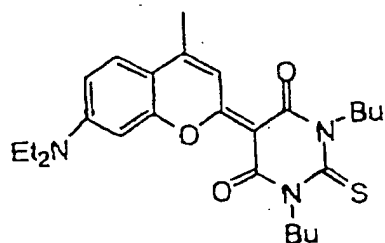
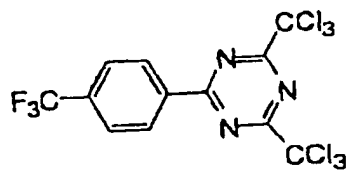
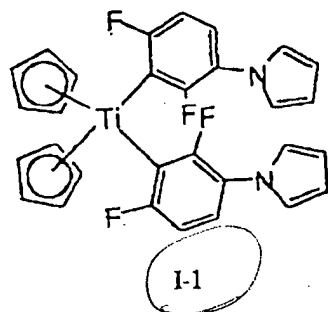
## [Addition-polymerizable Compounds]

**[0154]**

- M-1: Pentaerythritol tetraacrylate (NK Ester A-TMMT, manufactured by Shin-Nakamura Chemical Co., Ltd.)  
M-2: Glycerol dimethacrylate/hexamethylene diisocyanate urethane prepolymer (UA101A, manufactured by Kyoeisha Chemical Co., Ltd.)  
M-3: Dipentaerythritol acrylate (NK Ester A-9530, manufactured by Shin-Nakamura Chemical Co., Ltd.)

[Photopolymerization Initiator Materials in Tables 3 and 4]

[0155]



[Binder Polymers]

[0156]

PR-1:

Allyl methacrylate (POC-1)/methacrylic acid copolymer (comonomer proportion, 80/20 by mole)

Acid value determined by NaOH titration, 1.70 meq/g

Weight-average molecular weight determined by GPC, 48,000

PR-2:

Allyl methacrylate

(POC-1)/2-[2-(methacryloyloxy)ethoxycarbonyl]benzoic acid copolymer (comonomer proportion, 70/30 by mole)

Acid value determined by NaOH titration, 1.72 meq/g

Weight-average molecular weight determined by GPC, 95,000

PR-3:

Allyl methacrylate/3-[2-(methacryloyloxy)ethoxycarbonyl]-propionic acid copolymer (comonomer proportion, 70/30 by mole)

Acid value determined by NaOH titration, 1.72 meq/g

Weight-average molecular weight determined by GPC, 95,000

PA-1:

Polyurethane resin which is a condensation polymer of the following diisocyanates and diols

4,4'-Diphenylmethane diisocyanate (MDI)

Hexamethylene diisocyanate (HMDI)

Polypropylene glycol; weight-average molecular weight, 1,000 (PPG1000)

2,2-Bis(hydroxymethyl)propionic acid (DMPA)

Comonomer proportion (MDI/HMDI/PPG1000/DMPA), 40/10/15/35 by mole

Acid value determined by NaOH titration, 1.05 meq/g

Weight-average molecular weight determined by GPC, 45,000

PA-2:

Methyl methacrylate/acrylonitrile/N-[(4-sulfamoyl)phenyl]methacrylamide copolymer (comonomer proportion, 37/30/33 by mole)

Weight-average molecular weight determined by GPC, 49,000

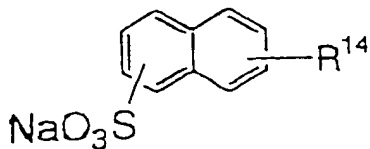
[Developing Solutions]

D-1:

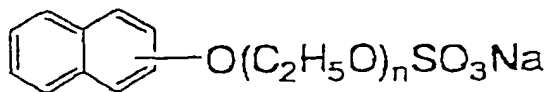
[0157] Aqueous solution of pH 10 having the following composition

Monoethanolamine	0.1 weight parts
Triethanolamine	1.5 weight parts
Compound of the following formula 1	4.0 weight parts
Compound of the following formula 2	2.5 weight parts
Compound of the following formula 3	0.2 weight parts
Water	91.7 weight parts

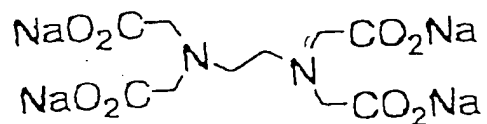
Formula 1:



Formula 2:



Formula 3:



[0158] In formula 1 given above, R<sup>14</sup> represents a hydrogen atom or butyl group.

D-2:

[0159] Aqueous solution having the following composition

1K potassium disilicate	3.0 weight parts
Potassium hydroxide	1.5 weight parts
Compound of the formula 3 given above	0.2 weight parts
Water	95.3 weight parts

TABLE 3

No.	Support	Photosensitive layer					Develop- ing solution	Printing durability Image Area (x10000 sheets)	Dot area
		Addition polymer- izable compound	Blinder polymer	Photo- polymer- ization initiator	Sensi- tizing dye	Co-sensi- tizing dye			
Ex. 1	AS-3	M-1	P-1	I-1	S-1	C-1	D-2	40	Excellent
Ex. 2	AS-3	M-1	P-2	I-1	S-2	C-1	D-2	22	Good
Ex. 3	AS-3	M-2	P-4	I-2	S-1	C-1	D-2	33	Excellent
Ex. 4	AS-1	M-2	P-5	I-1	S-2	C-1	D-1	35	Good
Ex. 5	AS-3	M-2	P-7	I-1	S-2	C-2	D-2	33	Good
Ex. 6	AS-3	M-3	P-9	I-2	S-1	C-3	D-2	27	Excellent
Ex. 7	AS-3	M-1	P-11	I-1	S-1	C-1	D-2	28	Good
Ex. 8	AS-3	M-1	P-13	I-1	S-2	C-1	D-1	37	Excellent
Ex. 9	AS-1	M-2	P-16	I-2	S-1	C-1	D-2	37	Excellent
Ex. 10	AS-1	M-2	P-17	I-2	S-1	C-2	D-1	34	Excellent
Ex. 11	AS-3	M-2	P-18	I-1	S-1	C-2	D-2	21	Good
Ex. 12	AS-3	M-2	P-20	I-1	S-2	C-1	D-2	27	Excellent
Ex. 13	AS-1	M-3	P-22	I-2	S-1	C-1	D-2	25	Excellent
Ex. 14	AS-3	M-3	P-23	I-1	S-2	C-3	D-1	23	Excellent
Ex. 15	AS-3	M-1	P-24	I-2	S-1	C-1	D-2	27	Good
Ex. 16	AS-3	M-2	P-25	I-1	S-2	C-1	D-2	27	Excellent
Ex. 17	AS-1	M-3	P-26	I-2	S-1	C-1	D-2	25	Good
Ex. 18	AS-3	M-3	P-27	I-1	S-2	C-3	D-1	23	Excellent
Ex. 19	AS-1	M-2	P-28	I-1	S-2	C-3	D-1	30	Good
Ex. 20	AS-3	M-1	P-29	I-2	S-1	C-1	D-2	27	Good
Comp. Ex. 1	AS-3	M-1	PR-1	I-1	S-1	C-1	D-2	5	Excellent
Comp. Ex. 2	AS-3	M-1	PR-2	I-2	S-2	C-1	D-2	11	Exist. of slipping
Comp. Ex. 3	AS-3	M-2	PR-3	I-1	S-1	C-2	D-2	4	Exist. of slipping



TABLE 4

No.	Support	Photosensitive layer						Develop- ing Solution	Image area (x 10000 sheets)	Printing durability Dot area	
		Addition polymeri- zable compound	Binder polymer		Weight ratio (1:2)	Photo- Polymeri- zation initia- tor	Sensi- tizing dye				Co-sens- itizing dye
			1	2							
Ex. 21	AS-1	M-1	P-3	P-6	5/5	I-1	S-2	C-1	1.4	30	good
Ex. 22	AS-3	M-3	P-8	P-21	6/4	I-1	S-1	C-1	1.2	28	good
Ex. 23	AS-3	M-2	P-8	P-30	7/3	I-2	S-1	C-2	1.2	34	good
Ex. 24	AS-1	M-2	P-10	P-21	8/2	I-1	S-1	C-1	1.4	33	good
Ex. 25	AS-3	M-1	P-12	P-28	9/1	I-2	S-2	C-3	1.4	21	excellent
Ex. 26	AS-3	M-1	P-14	P-15	3/7	I-2	S-1	C-1	1.6	24	excellent
Ex. 27	AS-1	M-2	P-1	PA-1	8/2	I-1	S-1	C-1	1.4	21	excellent
Ex. 28	AS-3	M-3	P-10	PA-2	5/5	I-1	S-1	C-1	1.4	23	excellent
Ex. 29	AS-3	M-2	P-10	PA-2	7/3	I-2	S-2	C-3	1.7	22	excellent
Ex. 30	AS-3	M-1	P-26	PR-3	7/3	I-2	S-2	C-1	1.2	27	good
Comp Ex. 4	AS-3	M-2	P-1	PA-1	5/5	I-2	S-2	C-1	1.3	9	Exist. of slipping
Comp Ex. 5	AS-1	M-2	P-1	PA-2	8/2	I-1	S-1	C-1	1.5	4	good
Comp Ex. 6	AS-3	M-3	P-3	PA-1	9/1	I-1	S-1	C-3	1.5	20	Exist. of slipping

EXAMPLES 31 TO 50 AND COMPARATIVE EXAMPLES 7 TO 9

## [Formation of Photosensitive Layer]

- 5 **[0160]** The coating solution for photosensitive layer formation shown below was prepared and applied to the aluminum support [AS-4] with a wire bar in the dry coating amount shown in Table 5. The coating was dried with a hot-air drying oven at 115°C for 45 seconds to form a photosensitive layer.

10	Coating Solution for Photosensitive layer:	
	Addition-polymerizable compound (compound shown in Table 5)	1.0 g
	Binder polymer (compound shown in Table 5)	1.0 g
	Infrared absorber (IR-1)	0.08 g
	Heat-polymerization initiator (compound shown in Table 5)	0.3 g
15	Fluorine-based nonionic surfactant (Megafac F-176, manufactured by Dainippon Ink & Chemicals Inc.)	0.01 g
	Victoria Pure Blue naphthalenesulfonate	0.04 g
	Methyl ethyl ketone	9.0 g
20	Propylene glycol monomethyl ether	8.0 g
	Methanol	10.0 g

## [Coating of Protective Layer]

- 25 **[0161]** A 3% by weight aqueous solution of a poly (vinyl alcohol) (degree of saponification, 98 mol%; degree of polymerization, 550) was applied according to need to the photosensitive layer in an amount of 2 g/m<sup>2</sup> in terms of dry coating amount. The coating was dried at 100°C for 2 minutes.

## [Exposure of Negative working Photosensitive Lithographic Printing Plates]

- 30 **[0162]** The negative working photosensitive lithographic printing plates thus obtained were exposed with Trendsetter 3244 VFS, manufactured by Creo and equipped with a water-cooled 40-W infrared semiconductor laser, under the conditions of an output of 9 W, external-drum rotational speed of 210 rpm, plate-surface energy of 100 mJ/cm<sup>2</sup>, and resolution of 2,400 dpi.

## 35 [Development/Platemaking]

- [0163]** After the exposure, the developing solution shown in Table 5 and a solution obtained by diluting finisher FN-6, manufactured by Fuji Photo Film Co., Ltd., with water in a ratio of 1:1 were introduced into automatic processor  
40 Stablon 900N, manufactured by Fuji Photo Film Co., Ltd. The plates exposed were developed at 30°C to obtain lithographic printing plates.

## [Image Area Printing Durability Test]

- 45 **[0164]** Printing machine Lithron, manufactured by Komori Corp., was used together with Graph G (N), manufactured by Dainippon Ink & Chemicals Inc., as an ink. The solid-image part of each printed matter was examined. The printing durability of the image areas was evaluated in terms of the number of printed sheets obtained by the time when the image began to be sharpened. The larger the number, the better the printing durability.

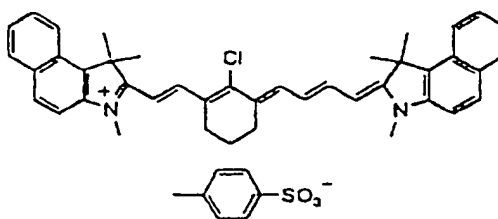
## 50 [Forced Dot Area Printing Durability Test]

- [0165]** Printing machine Lithron, manufactured by Komori Corp., was used together with Graph G (N), manufactured by Dainippon Ink & Chemicals Inc., as an ink. At the time when 5,000 sheets had been printed since printing initiation, the dot areas were wiped with a printing sponge impregnated with PS plate cleaner CL-2, manufactured by Fuji Photo  
55 Film Co., Ltd., to remove the ink adherent to the plate surface. Thereafter, 10,000 sheets were printed, and the printed matters obtained were visually examined for slipping in the dot images.

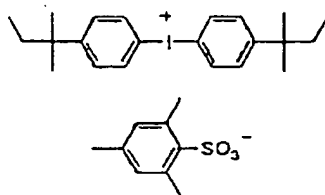
TABLE 5

No.	Support	Photosensitive layer				Protective layer	Developing Solution	Printing durability	
		Addition Polymerizable compound	Binder polymer	Weight ratio (1:2)	Heat-Polymerization initiator			Image area (x10000 sheets)	Dot area
Ex. 31	AS-4	M-3	P-2	-	OI-1	none	D-3	38	Excellent
Ex. 32	AS-4	M-3	P-3	-	OI-1	present	D-3	40	Good
Ex. 33	AS-1	M-2	P-5	-	OI-2	none	D-2	25	Good
Ex. 34	AS-4	M-1	P-7	-	OI-1	present	D-1	33	Excellent
Ex. 35	AS-2	M-3	P-11	-	OI-2	none	D-3	21	Excellent
Ex. 36	AS-4	M-3	P-14	-	OI-3	none	D-3	28	Excellent
Ex. 37	AS-3	M-2	P-16	-	OI-1	present	D-3	32	Excellent
Ex. 38	AS-3	M-2	P-19	-	OI-1	none	D-2	23	Excellent
Ex. 39	AS-4	M-1	P-21	-	OI-2	none	D-1	36	Excellent
Ex. 40	AS-4	M-1	P-23	-	OI-2	present	D-1	27	Good
Ex. 41	AS-3	M-3	P-25	-	OI-3	present	D-3	30	Good
Ex. 42	AS-3	M-3	P-27	-	OI-1	none	D-2	24	Good
Ex. 43	AS-4	M-3	P-29	-	OI-2	none	D-1	34	Good
Ex. 44	AS-4	M-2	P-10	8/2	OI-1	present	D-3	33	Good
Ex. 45	AS-2	M-1	P-12	9/1	OI-3	none	D-3	21	Excellent
Ex. 46	AS-4	M-3	P-14	8/2	OI-2	none	D-2	24	Excellent
Ex. 47	AS-4	M-2	P-1	8/2	OI-2	none	D-1	36	Excellent
Ex. 48	AS-3	M-3	P-10	5/5	OI-1	present	D-3	23	Excellent
Ex. 49	AS-3	M-2	P-10	7/3	OI-1	none	D-3	22	Excellent
Ex. 50	AS-1	M-3	P-26	7/3	OI-2	none	D-1	27	Good
Comp. Ex. 7	AS-4	M-3	P-1	-	OI-1	none	D-1	3	Exist. of slipping
Comp. Ex. 8	AS-4	M-2	P-2	-	OI-1	none	D-3	5	Good
Comp. Ex. 9	AS-4	M-3	P-3	9/1	OI-3	present	D-3	10	Exist. of slipping

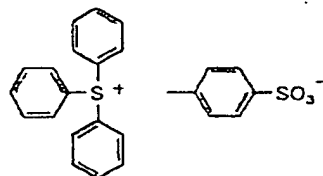
IR-1



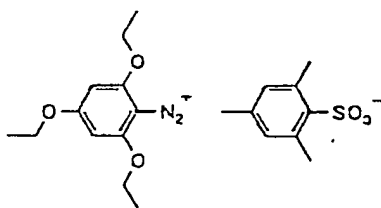
OI-1



OI-2



OI-3



[Developing Solution in Table 5]

D-3:

Sodium carbonate monohydrate	10 g
Potassium hydrogen carbonate	10 g
Sodium isopropylthiophthalenesulfonate	15 g
Sodium dibutylthiophthalenesulfonate	15 g
Sodium salt of ethylene glycol mononaphthyl ether monosulfate	10 g
Sodium sulfite	1 g
Tetrasodium ethylenediaminetetraacetate	0.1 g
Ion-exchanged water	938.9 g

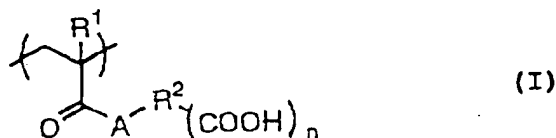
[0166] Tables 3 to 5 show that printing plates having exceedingly high printing durability were obtained from the negative working photosensitive lithographic printing plates having a photosensitive layer containing a polymeric binder having repeating units represented by formula (I).

[0167] As described above, the negative working photosensitive lithographic printing plate of the present invention gives a lithographic printing plate having extremely high printing durability because of the photosensitive layer thereof which contains a polymeric binder having repeating units represented by formula (I). The negative working photosensitive lithographic printing plate of the present invention is suitable for scanning exposure to a laser light, is capable of high-speed writing, and further attains high printing plate productivity.

[0168] While the invention has described in detail and with reference to specific embodiments thereof, it will be apparent to one skilled in the art that various changes and modifications can be made therein without departing from the spirit and scope thereof.

## Claims

1. A negative working photosensitive lithographic printing plate which comprises a support having thereon at least one photosensitive layer containing a polymeric binder having repeating units represented by formula (I):



wherein  $\text{R}^1$  represents a hydrogen atom or a methyl group;  $\text{R}^2$  represents a hydrocarbon group which has an alicyclic structure and has 3 to 30 carbon atoms and a valence of  $n+1$ ; A represents an oxygen atom or  $-\text{NR}^3-$ , wherein  $\text{R}^3$  represents a hydrogen atom or a monovalent hydrocarbon group having 1 to 10 carbon atoms; and  $n$  is an integer of 1 to 5.

2. The negative working photosensitive lithographic printing plate as in claim 1, wherein  $\text{R}^2$  is a hydrocarbon group which has an alicyclic structure comprising two or more rings and has 5 to 30 carbon atoms and a valence of  $n+1$ .
3. The negative working photosensitive lithographic printing plate as in claim 1, wherein the photosensitive layer further contains a photo- or heat-polymerization initiator and a compound having an ethylenically unsaturated bond capable of addition polymerization.



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# EUROPEAN SEARCH REPORT

Application Number  
EP 01 11 6791

DOCUMENTS CONSIDERED TO BE RELEVANT			
Category	Citation of document with indication, where appropriate, of relevant passages	Relevant to claim	CLASSIFICATION OF THE APPLICATION (Int.Cl.7)
X	JP 58 134629 A (MITSUBISHI CHEM. IND. LTD.) 10 August 1983 (1983-08-10) * page 159, column 17, line 15 - line 16 * * page 160, column 22 *	1-3	603F7/033 B41C1/10
A	JP 11 258801 A (FUJI PHOTO FILM CO., LTD.) 24 September 1999 (1999-09-24) * page 9 - page 17 *	1-3	
A, D	JP 2000 089469 A (KONICA CORPORATION) 31 March 2000 (2000-03-31) * page 5 *	1-3	
			TECHNICAL FIELDS SEARCHED (Int.Cl.7)
			603F B41C
The present search report has been drawn up for all claims			
Place of search THE HAGUE		Date of completion of the search 24 October 2001	Examiner Dupart, J.-M.
<p>CATEGORY OF CITED DOCUMENTS</p> <p>X : particularly relevant if taken alone Y : particularly relevant if combined with another document of the same category A : technological background O : non-written disclosure P : intermediate document</p> <p>T : theory or principle underlying the invention E : earlier patent document, but published on, or after the filing date D : document cited in the application L : document cited for other reasons &amp; : member of the same patent family, corresponding document</p>			

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EP 01 11 6791

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The members are as contained in the European Patent Office EDP file on  
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24-10-2001

Patent document cited in search report		Publication date	Patent family member(s)	Publication date
JP 58134629	A	10-08-1983	NONE	
JP 11258801	A	24-09-1999	NONE	
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